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Aliphatic alkanes and polycyclic aromatic hydrocarbons in atmospheric PM₁₀ aerosols from Baoji, China: Implications for coal burning

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ABSTRACT

Normal alkanes and PAHs in atmospheric PM₁₀ aerosols collected during 2008 winter and spring in Baoji, a mid-scale inland city of China, were determined on a molecular level. Concentrations of *n*-alkanes ranged from 232 to 3583 ng/m³ with an average of 1733 ng/m³ in winter and from 124 to 1160 ng/m³ with an average of 449 ng/m³ in spring, while PAHs in the PM₁₀ samples were 594 ± 405 and 128 ± 82 ng/m³ in the two seasons. Molecular compositions showed that CPI (odd/even) values of *n*-alkanes were close to unity for all the samples especially in winter, and diagnostic ratios of PAHs (e.g., Phe/(Phe + Ant), CPAH/ Σ PAHs and IcdP/(IcdP + BghiP)) were found similar to those in coal burning smoke with a strong linear relationship ($R^2 \ge 0.85$) between PAHs and fossil fuel derived *n*-alkanes, demonstrating that coal burning is the main source of *n*-alkanes and PAHs in the city, especially in winter due to house heating. Concentrations of the determined compounds in Baoji are much higher than those in Chinese mega-cities, suggesting that air pollution in small cities in the country is more serious and need more attention.

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1. Introduction

Coal burning smoke consists of a substantial fraction of carbonaceous aerosols. These aerosols are harmful to human health, because they contain some organics such as polycyclic aromatic hydrocarbons (PAHs), which are of great mutagenicity and carcinogenicity (Dockery et al., 1993). Moreover, the carbonaceous particles influence climate directly through scattering and adsorbing solar radiation and indirectly by acting as cloud condensation nuclei (CCN) (Ghan et al., 1995; Andreae, 2001; Menon et al., 2002). China is the biggest user of coal, who combusted 38.6% of the whole world coal consumption in 2006 (Lin, 2008). Around 25% of global anthropogenic carbonaceous particles are generated from China due to coal burning and petroleum combustion (Cooke et al., 1999), which could be transported with Asian dusts to other areas (Perry et al., 1999; Huebert et al., 2003; Tamamura et al., 2007), and probably influence air quality and climate in a regional and even a global scale.

Baoji (33°35′-35°06′N, 106°18′-108°03′E) is an inland city situated in the mid-west part of China with a population of 0.75 million. Coal usage in the city was 2.2 million tons in 2007, accounting for approximately 96% of the energy consumption (Baoji Statistic Yearbook, 2008). Many studies had focused on organic aerosols from coastal mega cities in China (Bi et al., 2003; Guo et al., 2003a; Guo et al., 2003b; Wang et al., 2006a; Hong et al., 2007), where economy is much more developed than inland China with higher efficiency and lower emissions for coal burning. However, the number (275) of the cities with similar population and economically developing levels as Baoji is much bigger than those mega cities (36), accounting for over 40% of the total number of cities in China (655) (Wu, 2004). Natural gas and liquefied petroleum gas used for house heating are not so common in the economically developing cities as those in the economically developed mega-cities. Instead, coal combustion is the main method for house heating in the mid-scale

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cities, which releases a huge amount of pollutants, leading to very serious air pollution in winter. However there has been no much attention paid for the problem in these mid-scale cities. To better understand the real situation of air pollution in China and its impact on regional and global climate change, it is very necessary to know the air pollution status in these cities. In the current study, we collected PM₁₀ samples at four locations of Baoji and performed a detailed characterization for the organic constituents to discuss the current status of air pollution in the city.

2. Experimental section

2.1. Sampling site

Four sampling sites were selected for aerosol collection. Environmental Monitoring Station (MS, 34°21′18.4″N, 107°08′34.7″E) represents the residential region, which is located in a big residential area; Da Qing Road (DQ, 34°21′ 52.5″N, 107°11′47.6″E) is a traffic center in Baoji; Miao Gou Village (MG, 34°17′24.0″N,107°11′20.0″E) is located far from city center, which represents the rural environment; Jing Yi Road (JY, 34°22′22.5″N, 107°08′34.1″E) is representative of the downtown area (Fig. 1).

2.2. Aerosol collection

Aerosols were collected by KC-6120 PM₁₀ samplers (Laoshan Electronic Instrument Company, Qingdao, China) equipped with pre-baked (500 °C, 4 h) quartz fiber filters (Whatman, QMA Ø90 mm). All the air inlets were 1.5 m above the ground and the flow rate was set as 100 L/min. Aerosol collection was conducted for 5/6 days in winter and spring, each started from 8:00 a.m. to 4:00 p.m. After sampling, the filter was sealed in an aluminum foil and stored at -18 °C prior to analysis. Meteorological conditions during the sampling periods were summarized in Table 1.



Fig. 1. Locations of four sampling sites: Environmental Monitoring Station (MS), Da Qing Road (DQ), Miao Gou Village (MG) and Jing Yi Road (JY) (http://earth.google.com/).

Table 1

Meteorological condition during sampling campaign.

| er ^a | 1, C | p, hPa | speed, m/s |
|-----------------|----------------|---------------------------|---|
| 60-87 | 2–12 | 953-961 | 0.7–1.5 |
| 57-81 | 16–27 | 933-943 | 1.3–1.9 |
| | 60-87 57-81 | 60-87 2-12 57-81 16-27 | 60-87 2-12 953-961 57-81 16-27 933-943 |

^a Sample number: sum of the four sites.

^b RH: relative humidity.

^c *p*: atmospheric pressure.

2.3. Sample extraction and derivatization

Extraction and derivatization procedures were described elsewhere (Wang and Kawamura, 2005), here we only give a brief introduction. One fourth of each filter was cut into pieces, extracted ultrasonically with dichloromethane/methanol (2:1, v/v) three times each for 10 min, and filtered through quartz wool packed in a Pasteur pipette. Afterward, the extracts were concentrated using a rotary evaporator under a vacuum condition and blown down to dryness with a gentle stream of pure nitrogen, then reacted with 50 µL of N, O-bis (trimethylsilyl) trifluoroacetamide (BSTFA) with 1% trimethyl chloride and 10 µL of pyridine to derivatize the compounds containing carboxyl and hydroxyl groups into the corresponding trimethylsilyl esters and ethers. Finally, 140 µL of hexane containing 1.43 ng/µL C₁₃ n-alkane as an internal standard was added to dilute the derivatives prior to determination. Here we only report *n*-alkanes and PAHs, and other components such as sugars and fatty acids will be reported later.

2.4. Gas chromatography-mass spectrometry quantification

All the derivatized extracts were quantified by Trace GC coupled with DSQ II mass spectrometer operated in electron impact mode (70 eV). The GC separation was carried out on a DB-5MS fused silica capillary column (30 m×0.25 mm i.d., 0.25 μ m film thickness) with the GC oven temperature programmed from 50 °C (2 min) to 120 °C at 15 °C min⁻¹ and then to 300 °C at 5 °C min⁻¹ with final isothermal hold at 300 °C for 16 min. The sample was injected on a splitless mode at an injector temperature of 280 °C. Identifications of individual compounds were conducted by comparing mass spectra with library data and authentic standards, and their GC-MS response factors were calculated based on the standards.

2.5. Quality assurance

Field blanks were collected at each site in both seasons, and treated by the same procedures for regular samples to determine any potential contamination. The results showed that the major contaminants were low molecular weight *n*-alkanes ($C_{15}-C_{21}$) and PAHs (MW=178, 202), of which the abundances were less than 5% of those in the real samples.

Additionally, 100–200 ng of the standards was spiked onto a clean filter and analyzed to determine their recoveries. The triplicate experiments showed that the average recoveries of n-alkanes, PAHs were in the range of 70–110%, except for the lighter *n*-alkanes (C_{15} - C_{18}), whose recoveries were lower than 70%. All the data reported here were corrected for field blanks but not for the recoveries.

3. Results and discussion

3.1. n-alkanes

3.1.1. General results

Concentrations of *n*-alkanes (C_{18} – C_{33}) of PM₁₀ in Baoji are listed in Table 2. The total concentrations varied from 232 to 3583 ng/m³ with an average of 1733 ng/m³ in the winter, and from 124 to 1160 ng/m³ with an average of 449 ng/m³ in the spring.

Lower molecular weight (LMW) n-alkanes (<C₂₃) are mainly derived from emissions of incomplete combustion of fossil fuels with a carbon preference index (CPI, a mass ratio of n-alkanes with odd carbon numbers to those with even carbon numbers)

close to unity, whereas high molecular weight (HMW) nalkanes $(>C_{25})$ are mostly derived from higher plant waxes with a CPI value far beyond 1.0 (Rogge et al., 1993a; Rogge et al., 1993b; Rogge et al., 1997; Schauer et al., 1999; Simoneit 2002; Cheng et al., 2006; Bi et al., 2008). In this study, concentrations of LMW *n*-alkanes (C_{18} - C_{25} , 1191 \pm 788 ng/m³) were much higher than those of HMW *n*-alkanes ($C_{26}-C_{33}$, 465 ± 259 ng/ m³) in winter, but an odd/even predominance could still be observed in the range of C_{26} – C_{33} , which might be resulted from the biomass burning in the city and/or other biogenic sources. CPI values in winter ranged from 1.1 to 1.5 with an average of 1.3, being similar to those in other Chinese cities such as Nanjing (1.2-1.4, average 1.3) (Wang and Kawamura, 2005) and Qingdao (1.2-1.8, average 1.4) (Guo et al., 2003b). In spring, abundances of the LMW *n*-alkanes $(231 \pm 126 \text{ ng/m}^3)$ were equivalent to the HMW ones $(218 \pm 156 \text{ ng/m}^3)$ with the highest concentration in the range of C₂₂–C₂₅. CPI values of the springtime *n*-alkanes were 1.6 ± 0.2 at the four sites, being

Table 2

Concentrations of *n*-alkanes and PAHs in PM_{10} during winter and spring in Baoji (ng/m³).

| Compound | Abbreviation | Winter (n: | =20) | | Spring $(n=24)$ | | | | |
|----------------------------|-----------------|------------|--|------|-----------------|------|---|------|------|
| | | Max | Min | Mean | Std | Max | Min | Mean | Std |
| I. n-alkanes | | | | | | | | | |
| Octadecane | C ₁₈ | 54.8 | 18.8 | 36.9 | 10.0 | 30.7 | 8.8 | 15.6 | 4.9 |
| Nonadecane | C ₁₉ | 62.8 | 9.3 | 33.9 | 15.7 | 18.4 | 4.5 | 9.7 | 3.2 |
| Eicosane | C ₂₀ | 188 | 11.3 | 85.7 | 55.0 | 22.6 | 5.2 | 12.5 | 5.4 |
| Heneicosane | C ₂₁ | 395 | 14.9 | 175 | 117 | 42.9 | 5.5 | 19.2 | 10.8 |
| Docosane | C ₂₂ | 505 | 24.4 | 241 | 153 | 53.2 | 7.6 | 25.1 | 15.8 |
| Tricosane | C ₂₃ | 568 | 27.8 | 266 | 171 | 95.4 | 10.6 | 46.7 | 28.7 |
| Tetracosane | C ₂₄ | 463 | 20.5 | 209 | 133 | 107 | 10.8 | 47.9 | 29.7 |
| Pentacosane | C ₂₅ | 429 | 24.1 | 200 | 127 | 126 | 13.3 | 54.5 | 31.7 |
| Hexacosane | C ₂₆ | 226 | 14.7 | 107 | 66.0 | 81.7 | 8.5 | 33.7 | 20.6 |
| Heptacosane | C ₂₇ | 206 | 15.2 | 102 | 56.4 | 139 | 14.3 | 51.4 | 33.6 |
| Octacosane | C ₂₈ | 90.7 | 9.7 | 44.2 | 24.9 | 60.4 | <lod< td=""><td>19.1</td><td>13.8</td></lod<> | 19.1 | 13.8 |
| Nonacosane | C ₂₉ | 192 | 22.9 | 103 | 42.1 | 148 | 10.7 | 47.9 | 32.2 |
| Tricontane | C ₃₀ | 65.6 | 3.0 | 23.0 | 15.0 | 41.7 | <lod< td=""><td>11.6</td><td>10.1</td></lod<> | 11.6 | 10.1 |
| Hentriacontane | C ₃₁ | 123 | 13.6 | 69.9 | 30.2 | 135 | <lod< td=""><td>33.5</td><td>28.6</td></lod<> | 33.5 | 28.6 |
| Dotriacontane | C ₃₂ | 35.5 | <lod< td=""><td>13.2</td><td>10.9</td><td>34.5</td><td><lod< td=""><td>7.4</td><td>9.4</td></lod<></td></lod<> | 13.2 | 10.9 | 34.5 | <lod< td=""><td>7.4</td><td>9.4</td></lod<> | 7.4 | 9.4 |
| Tritriacontane | C33 | 50.7 | <lod< td=""><td>23.1</td><td>14.7</td><td>57.5</td><td><lod< td=""><td>13.6</td><td>15.8</td></lod<></td></lod<> | 23.1 | 14.7 | 57.5 | <lod< td=""><td>13.6</td><td>15.8</td></lod<> | 13.6 | 15.8 |
| Subtotal | | 3583 | 232 | 1733 | 999 | 1160 | 124 | 449 | 268 |
| II. PAHs | | | | | | | | | |
| Phenanthrene | Phe | 75.3 | 10.5 | 38.0 | 15.7 | 20.2 | 3.9 | 10.2 | 4.1 |
| Anthracene | Ant | 9.1 | <lod< td=""><td>3.0</td><td>2.2</td><td>1.1</td><td><lod< td=""><td>0.5</td><td>0.3</td></lod<></td></lod<> | 3.0 | 2.2 | 1.1 | <lod< td=""><td>0.5</td><td>0.3</td></lod<> | 0.5 | 0.3 |
| Fluoranthene | Flu | 185 | 8.5 | 82.3 | 50.3 | 26.4 | 4.2 | 12.2 | 6.2 |
| Pyrene | Pyr | 119 | 6.6 | 56.8 | 34.9 | 19.0 | 4.2 | 10.7 | 4.5 |
| Benzolblfluorine | BF | 83.1 | 1.6 | 25.1 | 22.1 | 5.0 | 0.5 | 2.2 | 1.3 |
| Benzolalanthracene | BA | 123 | 1.8 | 39.5 | 33.5 | 11.0 | 0.5 | 5.1 | 3.7 |
| Chrysene/triphenylene | СТ | 140 | 3.4 | 52.3 | 37.2 | 28.3 | 1.7 | 10.5 | 7.1 |
| Benzolblfluoranthene | BbF | 290 | 5.8 | 107 | 76.2 | 84.6 | 3.8 | 27.7 | 20.3 |
| Benzolklfluoranthene | BkF | 68.2 | 1.2 | 25.7 | 19.0 | 23.8 | 0.7 | 7.5 | 5.7 |
| Benzolelpyrene | BeP | 82.4 | 1.8 | 31.3 | 21.6 | 29.7 | 1.2 | 9.3 | 6.8 |
| Benzolalpyrene | BaP | 68.2 | 1.3 | 24.6 | 19.0 | 15.8 | 0.6 | 5.7 | 4.0 |
| Pervlene | Per | 27.7 | <lod< td=""><td>8.2</td><td>7.0</td><td>5.2</td><td><lod< td=""><td>1.9</td><td>1.6</td></lod<></td></lod<> | 8.2 | 7.0 | 5.2 | <lod< td=""><td>1.9</td><td>1.6</td></lod<> | 1.9 | 1.6 |
| Indeno[1.2.3-cd]pyrene | IcdP | 91.6 | 1.9 | 34.6 | 25.2 | 24.2 | 1.2 | 8.3 | 6.1 |
| Dibenzola hlanthracene | DBA | 24.4 | <lod< td=""><td>68</td><td>6.4</td><td>61</td><td><1.0D</td><td>14</td><td>15</td></lod<> | 68 | 6.4 | 61 | <1.0D | 14 | 15 |
| Benzolghilpervlene | BohiP | 841 | 186 | 331 | 23.7 | 29.5 | 12 | 10.2 | 78 |
| Anthanthrene | Anta | 33.3 | <lod< td=""><td>7.9</td><td>9.4</td><td>4.4</td><td><lod< td=""><td>0.8</td><td>1.4</td></lod<></td></lod<> | 7.9 | 9.4 | 4.4 | <lod< td=""><td>0.8</td><td>1.4</td></lod<> | 0.8 | 1.4 |
| Coronene | Cor | 53.0 | <1.0D | 18.5 | 15.8 | 11.4 | <1.0D | 41 | 3.6 |
| Subtotal | | 1517 | 46 | 594 | 405 | 335 | 30 | 128 | 82 |
| Total, ng/m ³ | | 4908 | 278 | 2327 | 1383 | 1495 | 153 | 578 | 347 |
| $PM_{10} \mu g/m^3$ | | 574 | 207 | 402 | 100 | 691 | 158 | 410 | 160 |
| Total/PM ₁₀ , ‰ | | 11.2 | 1.4 | 5.4 | 2.6 | 2.5 | 0.6 | 1.4 | 0.5 |
| | | | | | | | | | |

LOD: limit of detection.

Table 3

| Concentration | and | diagnostic | ratio | of n- | alkanes | in | the | PM ₁₀ | samples | from | Baoji. |
|---------------|-----|------------|-------|-------|---------|----|-----|------------------|---------|------|--------|
| | | <u> </u> | | | | | | | | | |

| Location | Urban site | Rural site | | | | | | |
|--|--|--|---|--|--|---|--|---|
| | MS | | DQ | DQ | | | MG | |
| | Winter | Spring | Winter | Spring | Winter | Spring | Winter | Spring |
| \sum n-alkanes(ng/m ³) | 1598 ± 864 | 314 ± 199 | 1914 ± 539 | 552 ± 209 | 2828 ± 240 | 684 ± 240 | 592 ± 337 | 246 ± 91 |
| $C_n \max^a$ CPI ^b Plant wax C_n (%) ^c | $\begin{array}{c} C_{23} \\ 1.3 \pm 0.1 \\ 15.8 \pm 2.2 \end{array}$ | $\begin{array}{c} C_{25,\ 27,\ 29} \\ 1.7 \pm 0.2 \\ 27.4 \pm 5.4 \end{array}$ | $\begin{array}{c} C_{22,\ 23} \\ 1.2 \pm 0.1 \\ 10.6 \pm 2.2 \end{array}$ | $\begin{array}{c} C_{23,\ 24,\ 25} \\ 1.4 \pm 0.1 \\ 17.1 \pm 2.4 \end{array}$ | $\begin{array}{c} C_{23} \\ 1.3 \pm 0.1 \\ 14.9 \pm 2.8 \end{array}$ | $\begin{array}{c} C_{25,\ 27,\ 29,\ 31} \\ 1.6 \pm 0.2 \\ 23.5 \pm 4.4 \end{array}$ | $\begin{array}{c} C_{23,\ 29,\ 31} \\ 1.4 \pm 0.1 \\ 22.1 \pm 2.1 \end{array}$ | $\begin{array}{c} C_{27,\ 29} \\ 1.7 \pm 0.3 \\ 29.7 \pm 6.8 \end{array}$ |

^a C_n max: maximal peak of *n*-alkanes.

^bCPI: carbon preference index $(\sum C_{19}-C_{33}/\sum C_{18}-C_{32})$.

^cPlant wax *C_n*: calculated as the excess odd homologues – adjacent homologues average.

similar to those in Hong Kong (1.3–1.9, average 1.6) (Zheng et al., 2000) and Beijing (average 1.6) (Huang et al., 2006) but higher than those in Guangzhou (1.0–1.2, average 1.1) (Bi et al., 2003).

The molecular distribution pattern in winter suggests that emission from fossil fuel combustion is the major source of *n*alkanes in the Baoji atmosphere. Compared to those in spring the three times higher concentration of *n*-alkanes in the wintertime samples are most likely due to the sharp increase of coal burning for house heating, although the meteorological conditions in winter are favorable to accumulate the pollutants (Kadowaki, 1994).

3.1.2. Spatial and seasonal variations

At the downtown site JY, total concentrations of *n*-alkanes were significantly higher than those at other three sites with

an average of 2828 ± 633 and $684 \pm 240 \text{ ng/m}^3$ in winter and spring respectively, which is mainly caused by the emission from domestic coal burning of the restaurants without any emission control and the crowded buildings that prevent the dilution of the pollutants. CPI value at the site in winter is 1.3 ± 0.1 (Table 3) with the concentrations of *n*-alkanes peaking at C₂₃ (Fig. 2 a). In contrast, the molecular distributions of *n*-alkanes in spring were characterized by significant odd/even predominance in the range of C₂₆-C₃₃, although a slightly higher CPI value (1.6 ± 0.2) was found for the total homologues, indicating somewhat increase in the contribution from biological sources.

MS site, which is located in a residential area, had $1598 \pm$ 864 and $314 \pm 199 \text{ ng/m}^3$ of total *n*-alkanes in winter and spring separately. As seen in Fig. 2 b and Table 3, molecular



Fig. 2. Molecular distributions of *n*-alkanes in the PM₁₀ aerosols collected in winter (n=5) and spring (n=6) at the 4 sampling sites: (a) Jing Yi Road (JY), (b) Environmental Monitoring Station (MS), (c) Da Qing Road (DQ), and (d) Miao Gou Village (MG).

distributions, CPI values and relative abundances of plant wax derived *n*-alkanes were similar to JY site in both seasons, suggesting the similarity in the sources of the ambient particle associated *n*-alkanes at the two sites.

The second highest level of *n*-alkanes was obtained at site DQ, a traffic center in the city, with 1914 ± 539 and 552 ± 209 ng/m³ in winter and spring. During both seasons *n*-alkanes were dominated by LMW *n*-alkanes with a C_{max} at C_{22} - C_{25} (Fig. 2 c). CPI values were on average 1.2 ± 0.1 and 1.4 ± 0.1 in winter and spring with the lowest contribution from plant wax $(10.6\pm2.2\%$ in winter and 17.1 ± 2.5 in spring, Table 3) probably due to more contributions from traffic exhausts at the site.

MG is a background site and located in rural area about 8 km far away from the urban center. Concentrations of *n*-alkanes at MG were 592 ± 337 and 246 ± 91 ng/m³ in winter and spring with slightly higher levels of CPI values $(1.4 \pm 0.1 \text{ and } 1.7 \pm 0.3 \text{ in winter and spring, respectively, Table 3})$ and plant wax emissions $(22.1 \pm 2.1\% \text{ in winter and } 29.7 \pm 6.8\% \text{ in spring, Table 3})$. A bimodal distribution was found with two major peaks at C₂₃ and C₂₉ in winter and a unimodal distribution was observed in spring peaking at C₂₉ (Fig. 2d), suggesting the enhancement of springtime biogenic emissions.

3.2. PAHs

3.2.1. General results

Seventeen PAHs were detected for all the samples with Σ PAHs concentrations of $594 \pm 405 \text{ ng/m}^3$ in winter and $128 \pm 82 \text{ ng/m}^3$ in spring (Table 2). The decreased concentration in spring is largely due to the shutting down of coal burning for house heating. Benzo(b)flouranthene (BbF) was found to be the most abundant PAH in all the samples, accounting for 22% and 18% of the total PAHs in winter and spring, respectively, followed by fluoranthene (Flu), pyrene (Pyr) and chrysene/triphenylene (CT). Such a molecular distribution is consistent

with those reported by Wang et al. (2007) and Yang et al. (2005) for the samples from Nanjing in eastern China.

3, 4-Ring, 5-ring and 6,7-ring PAHs accounted for 50%, 33% and 17% of the total PAHs in winter and 40%, 41% and 19% of the total in spring with an increase in the relative abundance of wintertime 3,4-ring PAHs (Fig. 3). Most of LMW molecular PAHs (MW \leq 228) are present in the gas phase due to their high volatility (Jenkins et al., 1996; Bi et al., 2003; Chen et al., 2005), and tend to partition more into aerosols under a lower ambient temperature. Moreover, emission factors of these LMW PAHs are much higher than HMW PAHs for coal burning (Oros and Simoneit, 2000; Chen et al., 2005; Bi et al., 2008). Therefore, more LMW PAHs in the winter samples are probably resulted from the enhancement of coal burning and the lower temperature.

Though large-scale residential heating was stopped in spring, the total concentration of PM_{10} associated PAHs was still far beyond the levels in many other mega-cities in China (Wang et al., 2006b), which should primarily be attributed to the large amount of coal burned in the season with less emission control. In winter, total consumption of coal for residential heating was 0.35 million ton (Baoji Statistic Yearbook, 2008), accounting for about 13% of the annual usage, adding an additional source for PAHs in the air, which is associated with the lower temperature and calm wind, resulting in PAHs four times more abundant in winter (594 \pm 405 ng/m³, Table 2) than in spring (128 \pm 82 ng/m³).

3.2.2. Spatial and seasonal variations

Total PAHs presented the highest concentrations at the site JY and the lowest at the site MG in both seasons (Fig. 4), being consistent with the spatial distribution of *n*-alkanes. Unlike its isomer benzo(e)pyrene (BeP), benzo(a)pyrene (BaP) is liable to photochemical decomposition, thus ratio of BeP/(BeP + BaP) is an indicator for photochemical degradation (Nielsen, 1988; Spitzer and Kuwatsuka, 1993). The ratios $(0.57 \pm 0.05, \text{ Table 4})$ of BeP/(BeP + BaP) in winter were found to be lower than that (0.62 ± 0.04) in spring, indicating



Fig. 3. Relative abundance of PAHs with different molecular weight in winter and spring. (For compound abbreviations, see Table 2).



Fig. 4. Concentrations of total PAHs at the four locations.

a slight enhancement of photochemical oxidation in the warm season.

To further assess the origins of PAHs present in the samples, diagnostic ratios of PAHs that are relevant to the specific sources are calculated and showed in Table 4. Phe/ (Phe + Ant) and Flu/(Flu + Pyr) were usually taken as indicators of petroleum or combustion (Sicre et al., 1987; Budzinski et al., 1997). In this study, Phe/(Phe + Ant) ranged from 0.91 ± 0.01 to 0.98 ± 0.01 in winter and from 0.94 ± 0.01 to 0.97 ± 0.01 in spring at the four sites, being higher than the transition point of petroleum/combustion (0.90) (Budzinski et al., 1997) and implying a dominance of combustion. As reported by Yunker et al. (2002), Flu/(Flu + Pyr) ratios in the range of 0.40–0.50 and >0.50 represent liquid fossil fuel and solid fuel (i.e., grass, wood, and coal) combustions, respec-

tively. The average values for Flu/(Flu + Pyr) in this study were 0.59 ± 0.03 and 0.52 ± 0.04 in winter and spring, further suggesting a solid fuel combustion source in Baoji.

The nine major compounds (Flu, Pyr, BaA, Chry, BkF, BbF, BaP, IcdP, BghiP) are mainly produced through combustion processes (Prahl et al., 1984; Takada et al., 1990). The ratio of total concentration of the nine PAHs (CPAHs) to the whole (CPAHs/ \sum PAH) ranged from 0.30 to 0.51 for vehicle emissions (Rogge et al., 1993a). In this work, the average ratios were 0.76 and 0.75 during the two periods, indicating that particle associated PAHs in Baoji were less related to vehicle exhausts. The mean BghiP/BeP ratio was 1.06 in both seasons, which is far below the characteristic value (2.02 ± 0.17) for traffic emission but close to that (0.80) of non-traffic sources (Nielsen, 1996), again confirming the weak contribution from

Table 4

Diagnostic ratios of PAHs in the PM₁₀ samples from Baoji.

| Location | This study, Urban site | | | | | | | Rural site | | Source signature | | |
|----------------------------|------------------------|---------------|---------------|---------------|---------------|---------------|---------------|---------------|------------|---|---|------------------------------|
| | MS | | DQ | | ЈҮ | | MG | | combustion | Combustion sources | | Reference |
| | Winter | Spring | Winter | Spring | Winter | Spring | Winter | Spring | | Vehicle emission | Coal burning | |
| Phe/(Phe + Ant) | 0.92 ± 0.03 | 0.96 ± 0.02 | 0.93 ± 0.01 | 0.95 ± 0.01 | 0.91 ± 0.01 | 0.94±0.01 | 0.98 ± 0.01 | 0.97 ± 0.01 | <0.90 | >0.90 | | Budzinski et al., 1997 |
| Flu/(Flu+ Pyr) | 0.59 ± 0.02 | 0.50 ± 0.05 | 0.58 ± 0.01 | 0.52 ± 0.03 | 0.59 ± 0.01 | 0.54 ± 0.02 | 0.62 ± 0.04 | 0.53 ± 0.06 | <0.40 | 0.40- 0.50 | >0.50 | Yunker et al., 2002 |
| BeP/(BeP+ BaP) | 0.54 ± 0.05 | 0.59 ± 0.03 | 0.56 ± 0.04 | 0.62 ± 0.04 | 0.57 ± 0.04 | 0.64 ± 0.03 | 0.62 ± 0.03 | 0.62 ± 0.06 | | | | |
| CPAH∕ ∑PAH | 0.77 ± 0.01 | 0.77 ± 0.02 | 0.76 ± 0.01 | 0.75 ± 0.03 | 0.78 ± 0.01 | 0.77 ± 0.02 | 0.73 ± 0.03 | 0.73 ± 0.05 | | 0.30– 0.51 | | Rogge et al., 1993a |
| IcdP/ (IcdP + BghiP) | 0.51 ± 0.01 | 0.49 ± 0.03 | 0.50 ± 0.01 | 0.44 ± 0.04 | 0.52 ± 0.02 | 0.44 ± 0.02 | 0.50 ± 0.03 | 0.46 ± 0.04 | | | $\begin{array}{c} 0.51 \pm \\ 0.08 \end{array}$ | Bi et al., 2008 |
| BghiP/BeP | 1.13 ± 0.15 | 0.99 ± 0.28 | 1.06 ± 0.13 | 1.07 ± 0.12 | 1.04 ± 0.07 | 1.22 ± 0.12 | 1.02 ± 0.08 | 0.98 ± 0.13 | | $\begin{array}{c} 2.02 \pm \\ 0.17 \end{array}$ | | Nielsen, 1996 |



Fig. 5. Relationship of PAHs versus fossil fuels derived *n*-alkanes during the two seasons: (a) winter and (b) spring (ΣC_{PAHs} = total concentration of PAHs in each PM₁₀ sample; Σ fossil fuel C_n = Concentration of total *n*-alkanes – concentration of plant wax derived *n*-alkanes).

vehicle emissions. In contrast, IcdP/(BghiP + IcdP) was $0.51 \pm$ 0.02 in winter versus 0.46 ± 0.04 in spring, which is in agreement with the diagnostic ratio for coal burning smoke (0.51 ± 0.08) (Bi et al., 2008). Such molecular compositions of PAHs, therefore, clearly demonstrate that coal burning is the major source of PAHs in the city, which is consistent with that reported for the fourteen Chinese cities (Wang et al., 2006b). No significant differences in the PAHs ratios were observed among the four sites, suggesting the common source. Additionally, the seasonal variation of the diagnostic ratios was not obvious at the sites. Coal is the major energy source in Baoji throughout the whole year, and wintertime house heating is only one of many sections (e.g. power generation, metal smelting) that are dominated by coal burning, thus the diagnostic ratios of PAHs in spring are still characteristic of emissions of coal combustion.

3.2.3. Fossil fuel derived n-alkanes versus PAHs relationships

As mentioned in Section 3.1.1, high plant wax is an important source of n-alkanes, which could be calculated as

the excess odd homologues-adjacent even homologues average, and the difference from the total *n*-alkanes is the fossil fuel-derived amount (Simoneit et al., 1991). In this study, the fossil fuel derived *n*-alkanes contributed from 75.7% to 92.5% with an average of 84.2% in winter and from 60.1% to 86.7% with an average of 75.6% in spring. Additionally, a strong positive linear correlation was found between total PAHs and the fossil fuel derived *n*-alkanes in both seasons (Fig. 5 a and b), suggesting the similarity of sources for the two classes of compounds. According to the diagnostic ratios, combustion of coal is the dominant source of PAHs, thus most of the fossil fuel derived *n*-alkanes should also be ascribed to coal burning in the city.

3.3. The current situation and trend of atmospheric pollution in China

As shown in Table 5, the concentrations of n-alkanes and PAHs in the atmosphere of Chinese cities except Hong Kong are 1–2 orders of magnitude higher than those in other

Table 5

Comparison of *n*-alkanes and PAHs concentrations in Baoji with those in other international cities (ng/m^3) .

| City | Year | Size | Season | Reference | | | |
|-----------------------|-----------|---------------------|---------------|-------------------------|---------------|----------------|--------------------------------------|
| | | | Spring | Summer | Fall | Winter | |
| I. n-alkanes | | | | | | | |
| Baoji, China | 2008 | PM_{10} | $449\pm\!268$ | | | 1733 ± 999 | This study |
| Nanjing, China | 2001 | PM _{2.5} | | | 145(4-350) | 607(118-1407) | (Yang et al., 2005) |
| | 2004-2005 | PM _{2.5} | | 69 ± 32 | | 225 ± 113 | (Wang and Kawamura |
| | | | | | | | 2005) |
| Beijing, China | 2001-2002 | PM _{2.5} | 52 ± 19 | 39 ± 17 | 183 ± 130 | 477 ± 168 | (Huang et al., 2006) |
| | 2002 | PM _{2.5} | | 133(57–267) | | 514(112-1312) | (Feng et al., 2005) |
| | 2002-2003 | PM _{2.5} | | 51(24-78) | 333(143-749) | 375(102-651) | (He et al., 2006) |
| Guangzhou, China | 2001 | PM ₁₀ | 390 ± 262 | 71 ± 29 | | 0.54 | (Bi et al., 2002) |
| Qingdao, China | 2001-2002 | TSP | 73 | 53 | 90 | 371 | (Guo et al., 2003b) |
| Hong Kong, China | 1996-1997 | PIM _{2.5} | 20 ± 9 | 24 ± 11 | 26 ± 18 | 21 ± 5 | (Zheng et al., 2000) |
| Auckland, New Zealand | 1998-1999 | PIVI ₁₀ | 75(51-92) | 66(14-142) | 60(44-79) | 42(20 72) | (Yassaa et al., 2001) |
| Huston, USA | 1997-1998 | PIVI _{2.5} | 30(15-34) | 24(11-37) | | 42(26-72) | (Fraser et al., 2002) |
| II. PAHs | | | | | | | |
| This study | 2008 | PM_{10} | 128 ± 82 | | | 594 ± 405 | |
| Nanjing, China | 2001 | PM _{2.5} | | | 105(66-175) | 244(34-433) | (Yang et al., 2005) |
| | 2004-2005 | PM _{2.5} | | 40 ± 30 | | 87 ± 42 | (Wang et al., 2007) |
| Beijing, China | 2001-2002 | PM _{2.5} | 17 ± 8.0 | 8.4 ± 5.2 | 59 ± 52 | 229 ± 76 | (Huang et al., 2006) |
| | 2002 | PM _{2.5} | | 30(15-51) | | 212(33-547) | (Feng et al., 2005) |
| | 2002-2003 | PM _{2.5} | | 16(2.3-32) | 119(36-309) | 259(45-494) | (He et al., 2006) |
| | 2003-2005 | PM_{10} | 4 | 42 ± 32 Non-heating | ; season) | 305 ± 279 | (Okuda et al., 2006) |
| Guangzhou, China | 2001 | PM_{10} | 107 ± 44 | 19 ± 11 | | | (Bi et al., 2002) |
| | 2002-2003 | PM_{10} | 31(8.1-62) | 19(9.3-27) | 32(11-99) | 57(8.4-106) | (Tan et al., 2006) |
| | 2003-2004 | TSP | 32 ± 13 | 16 | 79 ± 12 | 84 ± 49 | (Duan et al., 2007) |
| Qingdao, China | 2001-2002 | TSP | 13 | 12 | 28 | 177 | (Guo et al., 2003b) |
| Hong Kong, China | 1996-1997 | PM _{2.5} | 2.5 ± 1.5 | 1.6 ± 0.4 | 2.1 ± 1.2 | 7.8 ± 3.5 | (Zheng et al., 2000) |
| | 2000-2001 | PM_{10} | | 5.3(1.9–10) | | 47(3.8–297) | (Guo et al., 2003a) |
| Tokyo, Japan | 1997 | TSP | | 14 ± 3.7 | | 36 ± 18 | (Tang et al., 2005) |
| Santiago, Brazil | 2000 | PM ₁₀ | 5.6(1.4-12) | | | 25(5.3–60) | (Del Rosario Sienra et al., 2005) |

regions/countries, among which the pollutants in Baoji are comparable and even more abundant than those in heavily polluted mega-cities. Such a severe air pollution problem in mid-scale cities like Baoji is mainly due to the increasing use of coal without strict emission controls.

Chinese government has made great efforts in dealing with the acid rain, planning a cut in SO₂ emission by 10% from 2005 to 2010. In the past three years, with the help of desulfurization before coal burning in power plant and other rigid controls, acid rain pollution in China has been mitigated with an decrease in total emission of SO₂ by 3.2%, but the air quality in more than 40% cities in the country are still in bad situation (China Environmental Bulletin, 2007). As seen in Table 5, no significant decrease could be found in the concentrations of n-alkanes and PAHs in Bejing and Guangzhou, and more serious pollution is observed in mid-scale cities like Baoji, thus a solution for carbonaceous aerosol pollution is in great need for now and the future.

4. Conclusion

In this study, 16 *n*-alkanes (C_{18} - C_{33}) and 17 PAHs of PM₁₀ in four sites of Baoji, a mid-scale city in China, were characterized on a molecular level with a GC/MS technique. Concentrations of *n*-alkanes were 1733 ± 990 ng/m³ for winter and 449 ± 268 ng/m³ for spring respectively, while concentrations of PAHs were 594 ± 405 and 128 ± 82 ng/m³ in both seasons. The much more abundant organic aerosols in winter are mainly due to the sharp increase in coal use for house heating. *n*-Alkanes and PAHs had similar spatial distributions in both seasons with the highest level at the downtown area and the lowest at the rural area. CPI values of *n*-alkanes, diagnostic ratios of PAHs and the strong correlation between the concentrations of PAHs and fossil fuel derived *n*-alkanes suggest that the PM₁₀ associated organics in the city are mainly originated from coal burning, especially in heating season. Compared to those in the mega cities, air pollution in mid-scale cities in China such as Baoji is much more serious and need more attention.

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References

Andreae, M.O., 2001. The dark side of aerosols. Nature 409, 671–672. Baoji Statistic Yearbook, 2008. Baoji Chorography Office, Baoji.

- Bi, X., Sheng, G., Peng, P.A., Zhang, Z., Fu, J., 2002. Extractable organic matter in PM10 from LiWan district of Guangzhou City, PR China. Science of the Total Environment 300, 213–228.
- Bi, X., Sheng, G., Peng, P.A., Chen, Y., Zhang, Z., Fu, J., 2003. Distribution of particulate- and vapor-phase n-alkanes and polycyclic aromatic hydrocarbons in urban atmosphere of Guangzhou, China. Atmospheric Environment 37, 289–298.

- Bi, X., Simoneit, B.R.T., Sheng, G., Fu, J., 2008. Characterization of molecular markers in smoke from residential coal combustion in China. Fuel 87, 112–119.
- Budzinski, H., Jones, I., Bellocq, J., Pierard, C., Garrigues, P., 1997. Evaluation of sediment contamination by polycyclic aromatic hydrocarbons in the Gironde estuary. Marine Chemistry 58, 85–97.
- Chen, Y., Sheng, G., Bi, X., Feng, Y., Mai, B., Fu, J., 2005. Emission factors for carbonaceous particles and polycyclic aromatic hydrocarbons from residential coal combustion in China. Environmental Science and Technology 39, 1861–1867.
- Cheng, Y., Li, S.-M., Leithead, A., Brook, J.R., 2006. Spatial and diurnal distributions of *n*-alkanes and *n*-alkan-2-ones on PM2.5 aerosols in the Lower Fraser Valley, Canada. Atmospheric Environment The Pacific 2001 Air Quality Study II, vol. 40, pp. 2706–2720.
- China Environmental Bulletin, 2007. Ministry of Environmental Protection of China, Beijing.
- Cooke, W.F., Liousse, C., Cachier, H., Feichter, J., 1999. Construction of a 1 degrees × 1 degrees fossil fuel emission data set for carbonaceous aerosol and implementation and radiative impact in the ECHAM4 model. Journal of Geophysical Research-Atmospheres 104, 22137–22162.
- Del Rosario Sienra, M., Rosazza, N.G., Préndez, M., 2005. Polycyclic aromatic hydrocarbons and their molecular diagnostic ratios in urban atmospheric respirable particulate matter. Atmospheric Research 75, 267–281.
- Dockery, D.W., Pope, C.A., Xu, X.P., Spengler, J.D., Ware, J.H., Fay, M.E., Ferris, B.G., Speizer, F.E., 1993. An association between air-pollution and mortality in 6 United States cities. New England Journal of Medicine 329, 1753–1759.
- Duan, J., Bi, X., Tan, J., Sheng, G., Fu, J., 2007. Seasonal variation on size distribution and concentration of PAHs in Guangzhou city, China. Chemosphere 67, 614–622.
- Feng, J., Chan, C.K., Fang, M., Hu, M., He, L., Tang, X., 2005. Impact of meteorology and energy structure on solvent extractable organic compounds of PM2.5 in Beijing, China. Chemosphere 61, 623–632.
- Fraser, M.P., Yue, Z.W., Tropp, R.J., Kohl, S.D., Chow, J.C., 2002. Molecular composition of organic fine particulate matter in Houston, TX. Atmospheric Environment 36, 5751–5758.
- Ghan, S.J., Chuang, C.C., Easter, R.C., Penner, J.E., 1995. A parameterization of cloud droplet nucleation .2. Multiple aerosol types. Atmospheric Research 36, 39–54.
- Guo, H., Lee, S.C., Ho, K.F., Wang, X.M., Zou, S.C., 2003a. Particle-associated polycyclic aromatic hydrocarbons in urban air of Hong Kong. Atmospheric Environment 37, 5307–5317.
- Guo, Z.G., Sheng, L.F., Feng, J.L., Fang, M., 2003b. Seasonal variation of solvent extractable organic compounds in the aerosols in Qingdao, China. Atmospheric Environment 37, 1825–1834.
- He, L.-Y., Hu, M., Huang, X.-F., Zhang, Y.-H., Tang, X.-Y., 2006. Seasonal pollution characteristics of organic compounds in atmospheric fine particles in Beijing. Science of the Total Environment 359, 167–176.
- Hong, H., Yin, H., Wang, X., Ye, C., 2007. Seasonal variation of PM10-bound PAHs in the atmosphere of Xiamen, China. Atmospheric Research 85, 429–441.
- Huang, X.-F., He, L.-Y., Hu, M., Zhang, Y.-H., 2006. Annual variation of particulate organic compounds in PM2.5 in the urban atmosphere of Beijing. Atmospheric Environment 40, 2449–2458.
- Huebert, B.J., Bates, T., Russell, P.B., Shi, G.Y., Kim, Y.J., Kawamura, K., Carmichael, G., Nakajima, T., 2003. An overview of ACE-Asia: strategies for quantifying the relationships between Asian aerosols and their climatic impacts. Journal of Geophysical Research-Atmospheres 108, 8633.
- Jenkins, B.M., Daniel, J.A., Turn, S.Q., Williams, R.B., 1996. Particle concentrations, gas-particle partitioning, and species intercorrelations for Polycyclic Aromatic Hydrocarbons (PAH) emitted during biomass burning. Atmospheric Environment 30, 3825–3835.
- Kadowaki, S., 1994. Characterization of carbonaceous aerosols in the Nagoya urban area. 2. Behavior and origin of particulate *n*-alkanes. Environmental Science and Technology 28, 129–135.
- Lin, B., 2008. China Energy Outlook. China Financial and Economic Press, Beijing, pp. 25–26.
- Menon, S., Hansen, J., Nazarenko, L., Luo, Y., 2002. Climate effects of black carbon aerosols in China and India. Science 297, 2250–2253.
- Nielsen, T., 1988. The decay of benzo(a)pyrene and cyclopenteno(cd)pyrene in the atmosphere. Atmospheric Environment (1967) 22, 2249–2254.
- Nielsen, T., 1996. Traffic contribution of polycyclic aromatic hydrocarbons in the center of a large city. Atmospheric Environment 30, 3481–3490.
- Okuda, T., Naoi, D., Tenmoku, M., Tanaka, S., He, K., Ma, Y., Yang, F., Lei, Y., Jia, Y., Zhang, D., 2006. Polycyclic aromatic hydrocarbons (PAHs) in the aerosol in Beijing, China, measured by aminopropylsilane chemically-bonded stationary-phase column chromatography and HPLC/fluorescence detection. Chemosphere 65, 427–435.
- Oros, D.R., Simoneit, B.R.T., 2000. Identification and emission rates of molecular tracers in coal smoke particulate matter. Fuel 79, 515–536.

- Perry, K.D., Cahill, T.A., Schnell, R.C., Harris, J.M., 1999. Long-range transport of anthropogenic aerosols to the National Oceanic and Atmospheric Administration baseline station at Mauna Loa Observatory, Hawaii. Journal of Geophysical Research-Atmospheres 104, 18521–18533.
- Prahl, F.G., Crecellus, E., Carpenter, R., 1984. Polycyclic aromatic-hydrocarbons in Washington coastal sediments – an evaluation of atmospheric and riverine routes of introduction. Environmental Science and Technology 18, 687–693.
- Rogge, W.F., Hildemann, L.M., Mazurek, M.A., Cass, G.R., Simoneit, B.R.T., 1993a. Sources of fine organic aerosol. 2. Noncatalyst and catalystequipped automobiles and heavy-duty diesel trucks. Environmental Science and Technology 27, 636–651.
- Rogge, W.F., Hildemann, L.M., Mazurek, M.A., Cass, G.R., Simoneit, B.R.T., 1993b. Sources of fine organic aerosol. 4. Particulate abrasion products from leaf surfaces of urban plants. Environmental Science and Technology 27, 2700–2711.
- Rogge, W.F., Hildemann, L.M., Mazurek, M.A., Cass, G.R., Simoneit, B.R.T., 1997. Sources of fine organic aerosol. 8. Boilers burning No. 2 distillate fuel oil. Environmental Science and Technology 31, 2731–2737.
- Schauer, J.J., Kleeman, M.J., Cass, G.R., Simoneit, B.R.T., 1999. Measurement of emissions from air pollution sources. 2. C1 through C30 organic compounds from medium duty diesel trucks. Environmental Science and Technology 33, 1578–1587.
- Sicre, M.A., Marty, J.C., Saliot, A., Aparicio, X., Grimalt, J., Albaiges, J., 1987. Aliphatic and aromatic hydrocarbons in different sized aerosols over the Mediterranean Sea: occurrence and origin. Atmospheric Environment (1967) 21, 2247–2259.
- Simoneit, B.R.T., Cardoso, J.N., Robinson, N., 1991. An assessment of terrestrial higher molecular weight lipid compounds in aerosol particulate matter over the south Atlantic from about 30–70°. Chemosphere 23, 447–465.
- Simoneit, B.R.T., 2002. Biomass burning a review of organic tracers for smoke from incomplete combustion. Applied Geochemistry 17, 129–162.
- Spitzer, T., Kuwatsuka, S., 1993. Residue levels of polynuclear aromaticcompounds in urban surface soil from Japan. Journal of Chromatography 643, 305–309.
- Takada, H., Onda, T., Ogura, N., 1990. Determination of polycyclic aromatichydrocarbons in urban street dusts and their source materials by capillary gas-chromatography. Environmental Science and Technology 24, 1179–1186.
- Tamamura, S., Sato, T., Ota, Y., Wang, X., Tang, N., Hayakawa, K., 2007. Longrange transport of polycyclic aromatic hydrocarbons (PAHs) from the eastern Asian continent to Kanazawa, Japan with Asian dust. Atmospheric Environment 41, 2580–2593.
- Tan, J.-H., Bi, X.-H., Duan, J.-C., Rahn, K.A., Sheng, G.-Y., Fu, J.-M., 2006. Seasonal variation of particulate polycyclic aromatic hydrocarbons associated with PM10 in Guangzhou, China. Atmospheric Research 80, 250–262.
- Tang, N., Hattori, T., Taga, R., Igarashi, K., Yang, X., Tamura, K., Kakimoto, H., Mishukov, V.F., Toriba, A., Kizu, R., Hayakawa, K., 2005. Polycyclic aromatic hydrocarbons and nitropolycyclic aromatic hydrocarbons in urban air particulates and their relationship to emission sources in the Pan-Japan Sea countries. Atmospheric Environment 39, 5817–5826.
- Wang, G., Kawamura, K., 2005. Molecular characteristics of urban organic aerosols from Nanjing: a case study of a mega-city in China. Environmental Science and Technology 39, 7430–7438.
- Wang, G., Huang, L., Zhao, X., Niu, H., Dai, Z., 2006a. Aliphatic and polycyclic aromatic hydrocarbons of atmospheric aerosols in five locations of Nanjing urban area, China. Atmospheric Research 81, 54–66.
- Wang, G., Kawamura, K., Lee, S., Ho, K.F., Cao, J.J., 2006b. Molecular, seasonal, and spatial distributions of organic aerosols from fourteen Chinese cities. Environmental Science and Technology 40, 4619–4625.
- Wang, G., Kawamura, K., Zhao, X., Li, Q., Dai, Z., Niu, H., 2007. Identification, abundance and seasonal variation of anthropogenic organic aerosols from a mega-city in China. Atmospheric Environment 41, 407–416.
- Wu, D., 2004. Demographics of the county and city in China. Department of the Ministry of Public Security, China. Qunzhong Press, Beijing, pp. 91–106.
- Yang, H., Yu, J.Z., Ho, S.S.H., Xu, J., Wu, W.-S., Wan, C.H., Wang, X., Wang, L., 2005. The chemical composition of inorganic and carbonaceous materials in PM2.5 in Nanjing, China. Atmospheric Environment 39, 3735–3749.
- Yassaa, N., Youcef Meklati, B., Cecinato, A., Marino, F., 2001. Particulate nalkanes, n-alkanoic acids and polycyclic aromatic hydrocarbons in the atmosphere of Algiers City Area. Atmospheric Environment 35, 1843–1851.
- Yunker, M.B., Macdonald, R.W., Vingarzan, R., Mitchell, R.H., Goyette, D., Sylvestre, S., 2002. PAHs in the Fraser River basin: a critical appraisal of PAH ratios as indicators of PAH source and composition. Organic Geochemistry 33, 489–515.
- Zheng, M., Fang, M., Wang, F., To, K.L., 2000. Characterization of the solvent extractable organic compounds in PM2.5 aerosols in Hong Kong. Atmospheric Environment 34, 2691–2702.