This article was downloaded by: [Institute of Earth Environment] On: 25 September 2012, At: 20:10 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of the Air & Waste Management Association

Publication details, including instructions for authors and subscription information: <u>http://www.tandfonline.com/loi/uawm20</u>

Winter and Summer PM_{2.5} Chemical Compositions in Fourteen Chinese Cities

Jun-Ji Cao^a, Zhen-Xing Shen^b, Judith C. Chow^{ac}, John G. Watson^{ac}, Shun-Cheng Lee^d, Xue-Xi Tie^{ae}, Kin-Fai Ho^a, Ge-Hui Wang^a & Yong-Ming Han^a

^a Key Lab of Aerosol, SKLLQG, Institute of Earth Environment, Chinese Academy of Sciences, Xi'an, China

^b Department of Environmental Sciences and Engineering, Xi'an Jiaotong University, Xi'an, China

^c Division of Atmospheric Sciences, Desert Research Institute, Reno, Nevada, USA

^d The Hong Kong Polytechnic University, Hong Kong

^e National Center for Atmospheric Research, Boulder, Colorado, USA

Accepted author version posted online: 24 Jul 2012. Version of record first published: 24 Sep 2012.

To cite this article: Jun-Ji Cao, Zhen-Xing Shen, Judith C. Chow, John G. Watson, Shun-Cheng Lee, Xue-Xi Tie, Kin-Fai Ho, Ge-Hui Wang & Yong-Ming Han (2012): Winter and Summer PM_{2.5} Chemical Compositions in Fourteen Chinese Cities, Journal of the Air & Waste Management Association, 62:10, 1214-1226

To link to this article: <u>http://dx.doi.org/10.1080/10962247.2012.701193</u>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <u>http://www.tandfonline.com/page/terms-and-conditions</u>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

TECHNICAL PAPER

Winter and Summer PM_{2.5} Chemical Compositions in Fourteen Chinese Cities

Jun-Ji Cao,^{1,*} Zhen-Xing Shen,² Judith C. Chow,^{1,3} John G. Watson,^{1,3} Shun-Cheng Lee,⁴ Xue-Xi Tie,^{1,5} Kin-Fai Ho,¹ Ge-Hui Wang,¹ and Yong-Ming Han¹

¹Key Lab of Aerosol, SKLLQG, Institute of Earth Environment, Chinese Academy of Sciences, Xi'an, China

²Department of Environmental Sciences and Engineering, Xi'an Jiaotong University, Xi'an, China

³Division of Atmospheric Sciences, Desert Research Institute, Reno, Nevada, USA

⁴The Hong Kong Polytechnic University, Hong Kong

⁵National Center for Atmospheric Research, Boulder, Colorado, USA

*Please address correspondence to: Jun-Ji Cao, Institute of Earth Environment, Chinese Academy of Sciences (CAS), No. 10 Fenghui South Road, High-Tech Zone, Xi 'an 710075, China; e-mail: cao@loess.llqg.ac.cn

 $PM_{2.5}$ in 14 of China's large cities achieves high concentrations in both winter and summer with averages >100 µg m⁻³ being common occurrences. A grand average of 115 µg m⁻³ was found for all cities, with a minimum of 27 µg m⁻³ measured at Qingdao during summer and a maximum of 356 µg m⁻³ at Xi'an during winter. Both primary and secondary $PM_{2.5}$ are important contributors at all of the cities and during both winter and summer. While ammonium sulfate is a large contributor during both seasons, ammonium nitrate contributions are much larger during winter. Lead levels are still high in several cities, reaching an average of 1.68 µg m⁻³ in Xi'an. High correlations of lead with arsenic and sulfate concentrations indicate that much of it derives from coal combustion, rather than leaded fuels, which were phased out by calendar year 2000. Although limited fugitive dust markers were available, scaling of iron by its ratios in source profiles shows ~20% of $PM_{2.5}$ deriving from fugitive dust in most of the cities. Multipollutant control strategies will be needed that address incomplete combustion of coal and biomass, engine exhaust, and fugitive dust, as well as sulfur dioxide, oxides of nitrogen, and ammonia gaseous precursors for ammonium sulfate and ammonium nitrate.

Implications: $PM_{2.5}$ mass and chemical composition show large contributions from carbon, sulfate, nitrate, ammonium, and fugitive dust during winter and summer and across fourteen large cities. Multipollutant control strategies will be needed that address both primary $PM_{2.5}$ emissions and gaseous precursors to attain China's recently adopted $PM_{2.5}$ national air quality standards.

Introduction

Suspended particulate matter (PM) is the major pollutant in many Chinese cities (Chan and Yao, 2008; Tie and Cao, 2009). Coal combustion to generate electricity and for domestic cooking and heating constitutes \sim 70% of the national energy budget (NAE et al., 2008). Total biomass burning in China, which includes domestic cooking and residential heating, field burning of crop residue, forest fires, and grassland fires, is estimated at 511.3 Tg yr⁻¹ (Yan et al., 2006). Improved engines and tighter emission standards are being offset by rapid growth in the motor vehicle fleet (Han and Hayashi, 2008). Paved and unpaved roads, construction, agricultural operations, and wind-blown soil eject geological material into the atmosphere (Du et al., 2008; Xuan et al., 2004). These and other emitters are contributing to high PM levels in Chinese cities, both through direct PM emissions and through conversion of sulfur dioxide (SO₂), nitrogen oxides (NO_x) , ammonia (NH_3) , and volatile organic compound (VOC)

gases to secondary sulfate (SO_4^{2-}) , nitrate (NO_3^{-}) , ammonium (NH_4^{+}) , and organic carbon (OC).

The Chinese government issued a national PM2.5 standard on February 29, 2012, that requires cities to have concentrations below 35 μ g m⁻³ annual average and <75 μ g m⁻³ for 24 hr, beginning in 2016 (http://cleanairinitiative.org/portal/node/ 8163). These standards were adopted owing to recognized adverse effects of PM_{2.5} chemical components on human health, visibility, and materials (Hu et al., 2009; Mauderly and Chow, 2008; Pope and Dockery, 2006; Watson, 2002). Elements, ions, and carbon fractions are often measured in PM2.5 to better evaluate the adverse effects and to indicate contributing sources. Several studies have reported these measurements in China (Cao et al., 2011; Chow et al., 2006; Deng et al., 2011; Duan et al., 2006; Gu et al., 2011; Guinot et al., 2007; He et al., 2001; Ho et al., 2006; Hu et al., 2010; Louie et al., 2005; Louie et al., 2005; Shen et al., 2007; So et al., 2007; Song et al., 2007; Sun et al., 2004; Wang et al., 2006; Wang et al., 2007; Wu et al., 2003; Xu et al., 2004; Yang et al., 2011; Zhang et al., 2010; Zhang and Friedlander, 2000; Zhao et al., 2010), but the areas studied, sampling site zones of representation, sampling periods, variables measured, and analysis methods are of insufficient consistency to evaluate similarities and differences. Reported here are consistently characterized simultaneous winter and summer PM_{2.5} mass and chemical concentrations obtained during 2003 at receptors with neighborhood and urban scale (Chow et al., 2002) in 14 of China's major cities. These measurements are used to compare and contrast the situation across a broad range of emissions and meteorology, examine seasonal changes, and assess contributions from coal combustion using elemental concentration ratios. These measurements from nearly a decade ago provide a baseline against which to evaluate future speciated PM_{2.5} measurements that will be needed to create and evaluate the multipollutant (Chow and Watson, 2011) control strategies required to attain the national standards.

Materials and Methods

As shown in Figure 1, measurement sites were located in 14 economically developed and developing cities across China. The neighborhood- and urban-scale sites were located on the campuses of schools and research institutes, as previously described (Cao et al., 2007; Cao et al., 2011; Han et al., 2009; Ho et al., 2007; Wang et al., 2006). Filter samplers were located on roof-tops at 6 to 20 m above ground level for around 2 weeks of sampling during winter (January 6–20) and summer (June 3–July 30) of 2003.

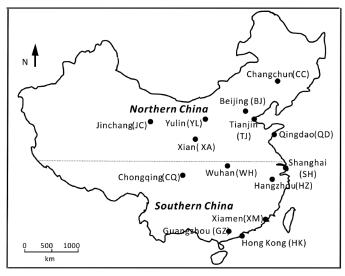


Figure 1. $PM_{2.5}$ samples were taken in seven southern China cities: Chongqing (CQ), Guangzhou (GZ), Hong Kong (HK), Hangzhou (HZ), Shanghai (SH), Wuhan (WH), and Xiamen (XM); and seven northern China cities: Beijing (BJ), Changchun (CC), Jinchang (JC), Qingdao (QD), Tianjin (TJ), Xi'an (XA), and Yulin (YL). Filter samples were obtained from 0900 to 0900 LST the next morning over 2-week periods during winter (January 6–20) and summer (June 3 – July 30) of 2003. Cities are classified as representing northern and southern China since: (1) precipitation events are more frequent and intense in southern China, and (2) northern China cities have lower wintertime temperatures, resulting in a greater amount of domestic heating, often using coal, along with shallower and more prolonged surface inversions at night and early morning.

PM_{2.5} samples were obtained on prefired (900°C, 3 h) 47-mm Whatman QM-A quartz-fiber filters by mini-volume air samplers (Airmetrics, Eugene, OR) at 5 L min⁻¹ flow rates. The exposed filters were stored at ~4°C after sampling, including shipping to the Xi'an laboratory, to minimize evaporation of volatile components. Filters were weighed before and after sampling with a ± 1 -µg sensitivity Sartorius MC5 electronic microbalance (Sartorius, Göttingen, Germany) after 24-hr equilibration at 20 to 23°C and 35 to 45% relative humidity (RH). Each filter was weighed at least three times before and after sampling. The maximum differences among the three repeated weights were less than 10 µg for blank filters and less than 20 µg for exposed filters. The collected PM was the difference between the average of exposed weights and the average of unexposed weights. Field blanks were also collected at each sampling site every seventh day by exposing filters in the sampler without drawing air through them; these were used to account for passive deposition or artefacts introduced between sample changing.

Elemental concentrations of Fe, Ti, Mn, Zn, As, Br, and Pb in filter deposits were determined by energy-dispersive x-ray fluorescence (ED-XRF) spectrometry (PANalytical Epsilon 5, Almelo, The Netherlands) (Chow and Watson, 2012; Watson et al., 2012). Other elements, such as Si, Ca, Al, and Mg, were not quantified owing to high and variable blank values on quartzfiber filters and potential biases caused by absorption of lowenergy x-rays from particles penetrating into the filter. XRF measurements on nine collocated Teflon-membrane and quartzfiber filters from Xi'an were comparable for these elements, with correlations (r) ranging from 0.982 for Fe and Zn (with slopes of 1.054 and 1.062, respectively) to 0.915 for As (with slope of 1.204). Measurement precision was determined as the standard deviation of several analyses of the same samples, yielding \pm 7.6% for Fe, \pm 8.6% for Ti, \pm 12.5% for Mn, \pm 7.6% for Zn, $\pm 23.5\%$ for As, $\pm 33.3\%$ for Br, and $\pm 7.9\%$ for Pb at typical concentration levels. Instrumental detection limits are 24.0 ng m^{-3} for Fe, 14.0 ng m^{-3} for Ti, 25.0 ng m^{-3} for Mn, 24.0 ng m^{-3} for Zn, 26.0 ng m⁻³ for As, 9.0 ng m⁻³ for Br, and 21.0 ng m⁻³ for Pb based on the uncertainties of blank filter counts. Replicate measurements were taken for every eight samples, and no differences were found that exceeded the precision intervals.

Following XRF analysis, the filter was sectioned with a precision cutter and one-fourth was extracted in 10 mL of distilled deionized water; the extract was submitted to ion chromatographic (IC) analysis (Shen et al., 2008; Shen et al., 2009) for cations Na⁺, NH₄⁺, and K⁺ and anions SO₄²⁻, NO₃⁻, and Cl⁻. Detection limits were 4.6 μ g L⁻¹ for Na⁺, 4.0 μ g L⁻¹ for NH₄⁺, 10.0 μ g L⁻¹ for K⁺, 0.5 μ g L⁻¹ for Cl⁻, 15 μ g L⁻¹ for NO₃⁻, and 20 μ g L⁻¹ for SO₄²⁻. Reference materials (National Research Center for Certified Reference Materials, China) agreed with analyses values within \pm 4%. One in 10 extracts was reanalyzed and none of the differences between these replicates exceeded precision intervals. Blank values were also subtracted from sample concentrations.

Organic carbon (OC) and elemental carbon (EC) were determined on a 0.5-cm² punch from each filter by a DRI model 2001 carbon analyzer (Atmoslytic, Inc., Calabasas, CA) following the IMPROVE thermal/optical reflectance (TOR) protocol (Cao et al., 2003; Chow et al., 1993; Chow et al., 2007; Chow et al., 2011). This produced four OC fractions (OC1, OC2, OC3, and OC4 at 120, 250, 450, and 550°C, respectively, in a helium [He] atmosphere); OP (a pyrolyzed carbon fraction determined when reflected laser light attained its original intensity after oxygen [O₂] was added to the analysis atmosphere); and three EC fractions (EC1, EC2, and EC3 at 550, 700, and 800°C, respectively, in a 2% $O_2/98\%$ He atmosphere). OC is defined as OC1 + OC2 + OC3 + OC4 + OP, and EC is defined as EC1 + EC2 + EC3 - OP.

Results and Discussion

PM_{2.5} mass concentrations

Figure 2 shows the wide distribution of concentrations observed across seasons and cities. The grand average of 115 μ g m⁻³ is more than 3 times the annual standard, and the highest 24-hr value of 543.9 μ g m⁻³, found in Xi'an during winter, is more than 7 times the 24-hr standard. OC, SO₄²⁻ NO₃⁻, NH₄⁺, and EC are the most abundant species, all with averages exceeding 5 μ g m⁻³. Elemental averages are less than the averages for carbon and ions, with Fe having the highest average of 2.4 μ g m⁻³ at Chongqing during winter. Concentrations ranged over several orders of magnitude, with the range increasing as the average concentration decreased. This variability indicates large spatial and temporal differences across the network.

Table 1 summarizes winter and summer $PM_{2.5}$ averages for each city. Standard deviations are typically 25% to 50% of the averages, indicating that these averages are not highly influenced by extreme events. Standard errors (standard deviation divided by the square root of the number of samples) of the averages are in the range of 6% to 13%.

In every city except Beijing and Xiamen (no summer data), wintertime $PM_{2.5}$ exceeded those of summertime, in many cases by a factor of 2 or more. Seasonal averages for $PM_{2.5}$ mass were similar in Beijing, with a winter/summer ratio of 0.88, in contrast

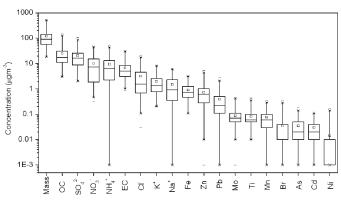


Figure 2. Average (square), median (central horizontal bar), 25th and 75th percentiles (lower and upper bars), 1st and 99th percentiles (lower and upper x), and minimum and maximum (—) concentrations for each chemical component across all cities and seasons. Average chemical components are ordered by abundance, with OC (24.5 μ g m⁻³), SO₄²⁻ (19.9 μ g m⁻³), NO₃⁻ (9.9 μ g m⁻³), NH₄⁺ (9.2 μ g m⁻³), EC (6.5 μ g m⁻³), Cl⁻ (3.1 μ g m⁻³), K⁺ (1.9 μ g m⁻³), and Na⁺ (1.5 μ g m⁻³) all being at important levels.

to the highest ratio of 4.9 at Qingdao, a coastal city in northern China. The lack of difference in the mass ratio for Beijing is partially due to the lack of change in the OC concentrations, which were $23.9 \pm 12.4 \,\mu g \,m^{-3}$ in winter, only 20% higher than summer. The winter/summer mass ratios for other cities are reflected in the major chemical component averages, which are 2 to 3 for OC, EC, and SO₄²⁻ in most cities, with NO₃⁻ and NH₄⁺ showing even higher winter/summer differences.

Average wintertime $PM_{2.5}$ was lowest in Xiamen (74.2 µg m⁻³) and highest in Xi'an (356.3 µg m⁻³). $PM_{2.5}$ was higher at inland cities (e.g., Xi'an, Wuhan, and Chongqing), and lower at the coastal (e.g., Xiamen and Hong Kong) and desert (i.e., Jinchang) cities. For the summer samples, average $PM_{2.5}$ was lowest in Qingdao (27.3 µg m⁻³), and highest in Beijing (131.6 µg m⁻³).

PM_{2.5} composition

OC and EC exhibited winter maxima and summer minima. OC was the most abundant wintertime $PM_{2.5}$ constituent in all cities except Hangzhou and Hong Kong, ranging from 13.3 (Hong Kong) to 95.8 µg m⁻³ (Xi'an). Wintertime EC levels vary with OC concentrations, which ranged from 4.6 (Jinchang) to 21.5 µg m⁻³ (Xi'an). This co-occurrence is expected, as OC and EC typically result from incomplete combustion of solid and liquid fuels (Lighty et al., 2000). OC and EC concentrations were highest in the inland cities, such as Changchun, Xi'an, Wuhan, and Chongqing, and lower in the coastal cities, such as Qingdao, Xiamen, and Hong Kong.

Wintertime SO_4^{2-} was the second most abundant component of $PM_{2.5}$ for all the cities except Hong Kong, varying from 11.5 μ g m⁻³ in Jinchang to 60.9 μ g m⁻³ in Chongqing. This was followed by NO₃⁻, ranging from 2.1 μ g m⁻³ (Jinchang) to 29 μ g m⁻³ (Xi'an), and NH₄⁺ ranged from 6.6 μ g m⁻³ (Jinchang) to 29.8 μ g m⁻³ (Xi'an). These high secondary ammonium sulfate ((NH₄)₂SO₄) and ammonium nitrate (NH₄NO₃) levels imply the need for precursor gas, as well as primary PM, emission reductions to reduce PM_{2.5} mass. The higher NH₄NO₃ values in winter than summer are consistent with a shift in equilibrium from the gas to particle phase with lower temperatures and higher RH (Stelson et al., 1979).

 K^+ is considered a marker for biomass burning (Andreae, 1983; Duan et al., 2004), although it is also a component of certain soils and sea spray (Pytkowicz and Kester, 1971). Wintertime K^+ levels exceeded 3 µg m⁻³ at Xi'an, Wuhan, Chongqing, and Hangzhou. The inland cities experience cold temperatures during winter and have abundant biomass available for residential heating.

Fe is a marker for fugitive dust, although it also originates from heavy industry. The wintertime Fe concentration was highest at 2.4 μ g m⁻³ in Chongqing, followed by Xi'an (1.8 μ g m⁻³), with the lowest wintertime average of 0.6 μ g m⁻³ at Xiamen. The two arid-region cities had low Fe concentrations, 1.2 μ g m⁻³ at Jinchang and 0.7 μ g m⁻³ at Yulin. The wintertime Fe averages did not correlate well with other soil components such as Ti and Mn across the sites, which may indicate additional Fe sources or variability in the fugitive dust compositions. Downloaded by [Institute of Earth Environment] at 20:10 25 September 2012

 3.7 ± 1.8 25.7 ± 14.8 33.4 ± 16.7 0.09 ± 0.02 0.13 ± 0.04 1.35 ± 0.31 1.35 ± 0.43 0.04 ± 0.00 0.08 ± 0.05 0.75 ± 0.17 0.27 ± 0.15 $0.2 \pm 0.09 \ 0.19 \pm 0.07 \ 1.75 \pm 0.52 \ 2.41 \pm 1.50 \ 0.11 \pm 0.04 \ 0.07 \pm 0.03 \ 1.68 \pm 0.60 \ 0.01 \pm 0.04 \ 0.07 \pm 0.03 \ 0.08 \pm 0.60 \ 0.01 \pm 0.00 \ 0.01 \pm 0.00 \ 0.01 \pm 0.00 \ 0.01 \pm 0.00 \ 0$ 0.28 ± 0.23 0.46 ± 0.09 0.16 ± 0.03 0.63 ± 0.28 0.07 ± 0.02 0.72 ± 0.26 0.42 ± 0.26 0.19 ± 0.09 0.48 ± 0.17 1.1 ± 0.48 0.6 ± 0.10 0.71 ± 0.31 0.02 ± 0.01 0.02 ± 0.01 0.22 ± 0.11 Ъ $0.17\pm0.05\ 0.08\pm0.03\ 1.25\pm0.32\ 1.33\pm0.46\ 0.09\pm0.02\ 0.01\pm0.01$ $0.09 \pm 0.02 \ 0.09 \pm 0.05 \ 0.95 \pm 0.28 \ 0.43 \pm 0.23 \ 0.02 \pm 0.01 \ 0.17 \pm 0.12$ $2.5 \pm 1.0 \quad 11.4 \pm 5.6 \quad 25.5 \pm 10.3 \quad 32.5 \pm 15.1 \quad 0.11 \pm 0.03 \quad 0.17 \pm 0.06 \quad 1.55 \pm 0.74 \quad 1.34 \pm 0.50 \quad 0.06 \pm 0.02 \quad 0.08 \pm 0.07 \quad 0.08 \pm 0.08 \quad 0.08 \quad 0.08 \pm 0.08 \quad 0.0$ $60.9 \pm 19.6 \ 0.15 \pm 0.02 \ 0.18 \pm 0.07 \ 2.41 \pm 0.75 \ 0.92 \pm 0.29 \ 0.08 \pm 0.03 \ 0.14 \pm 0.07$ $21.4 \pm 5.6 \quad 0.05 \pm 0.03 \quad 0.04 \pm 0.03 \quad 0.62 \pm 0.15 \quad 0.35 \pm 0.19 \quad 0.01 \pm 0.01 \pm 0.01 \\ \pm 0.01$ $4.1 \pm 1.2 \ \ 22.2 \pm 10.7 \ \ 31.4 \pm 15.6 \ \ 0.08 \pm 0.03 \ \ 0.11 \pm 0.05 \ \ 1.56 \pm 0.58 \ \ 1.11 \pm 0.48 \ \ 0.08 \pm 0.03 \ \ 0.03 \pm 0.02 \ \ 0.0$ $0.39 \pm 0.33 \ 0.04 \pm 0.04 \ 0.01 \pm 0.01$ $0.15\pm0.09\ 0.03\pm0.02\ 1.19\pm0.62\ 0.11\pm0.02\ 0.05\pm0.01\ 0.01\pm0.01$ 0 ± 0.01 $20.6 \pm 10.8 \ 0.06 \pm 0.03 \ 0.06 \pm 0.04 \ 0.78 \pm 0.51 \ 0.59 \pm 0.32 \ 0.04 \pm 0.02 \ 0.06 \pm 0.04$ $21.6 \pm 12.3 \ 0.09 \pm 0.02 \ 0.18 \pm 0.07 \ 1.18 \pm 0.38 \ 0.93 \pm 0.39 \ 0.03 \pm 0.01 \ 0.05 \pm 0.01$ Br $0.04 \pm 0.01 \ 0.03 \pm 0.01 \ 0.66 \pm 0.27 \ 0.05 \pm 0.03 \ 0.01 \pm 0.00$ \mathbf{As} Zn $0.09\pm0.03\ 0.07\pm0.03\ 0.99\pm0.32$ Е $0.06\pm 0.02\ 0.05\pm 0.01$ Mn Ξ (June 6–20, 2003) $9.5\pm4.4\ 29.0\pm10.0\ 53.8\pm25.6$ 20.0 ± 4.2 17.0 ± 5.1 20.2 ± 7.5 11.6 ± 2.6 21.1 ± 7.7 13.5 ± 8.4 SO_{4}^{2-} $4.5\pm2.5\ 11.7\pm4.1$ $1.3\pm1.5\ 10.9\pm4.4$ 11.5 ± 7.8 2.1 ± 1.4 $6.5\pm2.2\ 19.3\pm9.2$ $6.2\pm2.4\ 10.8\pm4.7\ 18.1\pm6.4$ 9.5 ± 2.7 $6.6\pm3.1\ 17.5\pm8.7$ 7.6 ± 4.1 13.1 ± 4.5 NO₃ 2.3 ± 1.8 1.9 ± 2.0 4.3 ± 1.3 1.2 ± 1.5 1.4 ± 0.5 5 1.1 ± 0.3 2.8 ± 1.3 $1.9\pm0.7\ 29.8\pm11.5\ 4.9\pm1.3$ 1.8 ± 1.2 0.8 ± 0.4 $1.2 \pm 0.4 \ 19.1 \pm 10.7 \ 4.4 \pm 1.3$ $1.7\pm0.7\ 18.4\pm10.2\ 3.4\pm1.3$ 0.6 ± 0.4 0.8 ± 0.3 1.4 ± 0.5 0.9 ± 0.5 2.1 ± 1.1 \mathbf{k}^+ $3.8\pm0.8\ 10.6\pm3.5$ 9.4 ± 4.1 9.4 ± 2.1 $2.9\pm 2.0\ 22.2\pm 9.8$ 8.0 ± 3.5 6.6 ± 1.4 $3.2\pm1.6\ 15.3\pm5.2$ $1.5\pm1.0\ 28.8\pm8.9$ 14.5 ± 5.9 7.0 ± 3.9 8.5 ± 7.6 NH4 1.4 ± 0.8 1.4 ± 0.4 1.5 ± 1.6 0.6 ± 0.5 1.9 ± 1.3 0.8 ± 0.6 0.4 ± 0.4 Na^+ 6.2 ± 2.7 13.4 ± 3.2 4.6 ± 1.1 6.2 ± 2.5 8.9 ± 2.5 $356.3 \pm 118.4 \ 95.8 \pm 27.7 \ 21.5 \pm 5.9$ 8.3 ± 3.0 $316.6\pm101.2\ 75.2\pm21.9\ 17.2\pm5.0$ 8.4 ± 6.6 6.9 ± 2.8 9.1 ± 2.1 8.6 ± 2.0 8.0 ± 3.4 4.9 ± 1.5 B 40.4 ± 7.04 24.2 ± 22.1 26.3 ± 10.3 43.0 ± 16.1 33.8 ± 15.8 23.9 ± 12.4 31.3 ± 8.24 17.6 ± 5.4 $88.37\pm23.14\ 13.3\pm5.1$ 26.7 ± 7.7 15.5 ± 5.7 30.5 ± 8.1 8 115.6 ± 46.6 134.8 ± 43.0 110.2 ± 79.6 177.3 ± 59.5 72.3 ± 67.0 148.1 ± 21.3 86.8 ± 14.2 203.1 ± 76.2 141.3 ± 56.0 139.4 ± 50.6 74.2 ± 27.2 Winter Mass Hong Kong Changchun Guangzhou Chongqing Hangzhou Shanghai Jinchang Qingdao Beijing (0)Tianjin (XA) $\widehat{\mathbf{S}}$ (HK) (HM) Xiamen Wuhan (MM) 00 (JL) (GZ) (HZ) (SH) (\mathbf{j}) (E) Xi'an Yulin (\mathbf{B}) City

Table 1. Arithmetic averages \pm standard deviations ($\mu g m^{-3}$) for PM_{2.5} mass and chemical components by city and season. See Figure 1 for city codes. Each average contains ~14 values

(Continued)

Downloaded by [Institute of Earth Environment] at 20:10 25 September 2012

Table 1. (Cont.)	ont.)															
City	Mass	OC	EC	Na^+	$\mathrm{NH_4}^+$	\mathbf{K}^{+}	CI^-	NO_3^-	$\mathrm{SO_4}^{2-}$	Ті	Mn	Fe	Zn	\mathbf{As}	Br	Pb
	Summer							()	(June 3–July 30, 2003)	3, 2003)						
Beijing (BJ)	131.6 ± 28.0	19.7±4.7	5.7 ± 4.1	0.5 ± 0.3	9.8 ± 4.2	2.5 ± 0.6	1.3 ± 0.6	13.7 ± 6.4	22.6 ± 9.2	$0.06\pm 0.02\ 0.06\pm 0.02$	0.06 ± 0.02	0.8 ± 0.24 0	$\pm \ 0.24 \ 0.44 \pm 0.15 \ 0.02 \pm 0.01$		$0.02\pm0.02\ 0$	0.18 ± 0.09
Changchun (CC)	51.0 ± 19.8	10.6 ± 2.4	2.6 ± 0.8	0.2 ± 0.1	2.1 ± 2.5	0.9 ± 0.3	0.6 ± 0.2	2.7 ± 1.0	8.4 ± 6.6	0.05 ± 0.02 (0.04 ± 0.02	0.55 ± 0.24	$0.5\pm0.20\ 0.01\pm0.00\ 0.01\pm0.00$	$.01 \pm 0.00$ ($0.01 \pm 0.00 0.01$	0.07 ± 0.06
Jinchang (JC)	49.7 ± 2.3	7.1 ± 0.9	1.6 ± 0.2	3.6 ± 0.1	0.4 ± 0.5	0.7 ± 0.2	1.0 ± 0.2	0.9 ± 0.2	7.8 ± 2.5	0.25 ± 0.27 (0.02 ± 0.02	0.58 ± 0.02 (0.15 ± 0.11 0	0.05 ± 0.03 C	0.01 ± 0.00	0.8 ± 0.88
Qingdao (OD)	27.3 ± 11.0	5.1 ± 3.1	1.5 ± 0.7	0.2 ± 0.4	1.6 ± 1.4	0.4 ± 0.2	0.6 ± 0.7	1.8 ± 1.0	6.7 ± 3.1	$0.03 \pm 0.03 \ 0.01 \pm 0.01$		0.26 ± 0.13 (0.03 ± 0.01	0 ± 0.00 C	$0\pm 0.00\ 0.01\pm 0.00\ 0.$	0.03 ± 0.02
Tianjin (TJ)	101.7 ± 26.2	16.4 ± 3.9	3.8 ± 1.5	3.5 ± 1.7	7.7 ± 3.8	1.3 ± 0.5	3.8 ± 2.3	8.7 ± 3.4	22.1 ± 9.1	0.05 ± 0.02 (0.05 ± 0.03 0	0.76 ± 0.31	$0.6\pm0.24\ 0.01\pm0.01$		0.04 ± 0.03 0	0.24 ± 0.13
Xi'an (XA)	102.8 ± 32.9	24.5 ± 6.7	6.5 ± 1.4	2.5 ± 1.9	4.3 ± 3.9	1.7 ± 0.4	1.1 ± 0.5	5.1 ± 2.8	16.8 ± 10.0	0.1 ± 0.03 (0.07 ± 0.03 1	1.08 ± 0.48 1	$1.22\pm0.62\ 0$	0.05 ± 0.04 C	0.01 ± 0.00 0.	0.75 ± 0.49
Yulin (YL)	45.6 ± 8.5	10.6 ± 1.7	3.4 ± 1.3	4.1 ± 0.5	0.7 ± 1.0	0.5 ± 0.1	0.8 ± 0.2	1.5 ± 0.9	9.0 ± 3.9	$0.06\pm 0.04\ 0.02\pm 0.01$		0.41 ± 0.12 0	0.07 ± 0.02 0	0.01 ± 0.01 C	0.01 ± 0.00 0	0.07 ± 0.03
Chongqing (CQ)	106.9 ± 40.4	23.7 ± 8.2	7.4 ± 1.6	0.8 ± 0.3	7.0 ± 3.6	2.3 ± 0.7	2.2 ± 1.4	4.2 ± 3.0	20.4 ± 9.4	0.07 ± 0.03	0.1 ± 0.06 0	$0.79\pm 0.40\ 0.67\pm 0.25$		0.03 ± 0.01 C	0.04 ± 0.02	0.2 ± 0.07
Guangzhou (GZ)	39.7 ± 6.6	9.9 ± 1.3	3.3 ± 0.5	0.6 ± 0.8	1.2 ± 0.8	1.0 ± 0.5	0.8 ± 0.7	1.2 ± 0.4	6.4 ± 1.5	0.06 ± 0.01 (0.02 ± 0.01	$.26 \pm 0.07$ ($0.26\pm0.07\ 0.66\pm0.16\ 0.02\pm0.01$		0.01 ± 0.01 0.	0.12 ± 0.05
Hong Kong (HK)	30.4 ± 7.3	6.6 ± 1.9	3.3 ± 1.1	$3.3 \pm 1.1 \ 0.72 \pm 0.3$	0.3 ± 0.2	0.6 ± 0.1	1.2 ± 0.4	1.2 ± 0.5	4.3 ± 1.2	$0.04\pm 0.02\ 0.04\pm 0.04$	$0.04 \pm 0.04 0$	$.27 \pm 0.11$ 0	$0.19 \pm 0.15 0$	$.01 \pm 0.00$ C	$0.27\pm0.11\ 0.19\pm0.15\ 0.01\pm0.00\ 0.01\pm0.00\ 0.01$	01 ± 0.02
Hangzhou (HZ)	80.0 ± 40.3	15.2 ± 3.5	3.2 ± 1.2	0.3 ± 0.4	5.3 ± 5.4	2.9 ± 1.1	0.6 ± 0.3	5.5 ± 7.8	16.5 ± 9.4	$0.06\pm 0.01 \ 0.05\pm 0.02$		0.51 ± 0.14 1	$1.12\pm0.26\ 0$	0.02 ± 0.01 C	$0.01 \pm 0.01 = 0.01$	0.33 ± 0.17
Shanghai (SH)	54.0 ± 23.1	13.4 ± 5.4	3.3 ± 1.9	0.6 ± 0.1	3.5 ± 2.5	0.9 ± 0.4	0.6 ± 0.2	2.6 ± 2.1	12.0 ± 5.7	0.06 ± 0.02 (0.06 ± 0.02	0.46 ± 0.28 (0.78 ± 0.23 0	0.02 ± 0.01 C	$0.02 \pm 0.01 0.01$	0.16 ± 0.11
Wuhan (WH)	70.7 ± 19.9	14.1 ± 3.7	2.7 ± 0.6	0.2 ± 0.2	2.9 ± 2.0	2.3 ± 1.2	0.5 ± 0.1	2.7 ± 1.0	12.4 ± 7.1	$0.05\pm 0.01 \ 0.08\pm 0.07$		$.93 \pm 0.28$ C	0.69 ± 0.32	$.05\pm0.04$ C	$0.93 \pm 0.28 \ 0.69 \pm 0.32 \ 0.05 \pm 0.04 \ 0.01 \pm 0.00 \ 0.36 \pm 0.25$	36 ± 0.25
Xiamen (XM)	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA

Note: NA, data not available.

The highest wintertime As $(0.11 \ \mu g \ m^{-3})$ and Pb $(1.7 \ \mu g \ m^{-3})$ concentrations were found at Xi'an. As and Pb are found in Chinese coal (Tian et al., 2011; Want et al., 2006), while Pb gasoline additives were discontinued in 2000 (Xu et al., 2012). The highest Br average was found at a coastal city, Qingdao (0.17 $\ \mu g \ m^{-3}$), consistent with a potential marine aerosol contribution.

Summertime averages were lower than those for winter for nearly all chemical components. In most cases, this can be attributed to warmer weather that improved dispersion and shifted the NH₄NO₃ from the particle to gas phase. Lower OC and EC averages are probably less related to domestic biomass and coal combustion, which is consistent with lower K⁺ and As averages. Engine exhaust and agricultural burning emissions are expected to contribute larger portions of OC and EC during summer.

 NO_3^- and NH_4^+ show the biggest contrast between winter and summer, consistent with the change in equilibrium. SO_4^{2-} levels were also much lower during summer than winter. This would be consistent with more nearby SO_2 to SO_4^{2-} conversion during winter, possibly in conjunction with reactive fogs and clouds (Pandis et al., 1992) and with local accumulation under stagnant conditions. The summer values could be more influenced by standard photochemical mechanisms occurring during long-range transport (Qian et al., 2001).

The Fe and Ti fugitive dust markers do not show a clear winter/summer pattern, being higher in some cities during summer and lower in others. The sampling periods did not include the April/May Asian dust storms (Gong and Zhang, 2008; Li et al., 2008) that are causes of high $PM_{2.5}$ during these periods. The other elements do not show major or consistent differences between winter and summer, except that the summertime averages are generally lower. The summer Pb average in Xi'an decreased by more than a factor of two (0.75 µg m⁻³).

Chemical ratios as source indicators

Several potential sources of different chemical components were mentioned earlier. These can be better understood by examining some of the elemental ratios available from the data set that might correspond to similar ratios in the source profiles. OC/EC ratios across the 14 cities are compared in Cao et al. (2007). Given the large role of domestic and industrial coal use, the SO_4^{2-}/OC , SO_4^{2-}/EC , NO_3^{-}/SO_4^{2-} , As/Fe, and Pb/Fe ratios are compared with ratios from other cities in Table 2. The 2003 SO_4^{2-}/OC ratio found in this study (0.90 \pm 0.43) is

much higher than that for the other cities, as is the SO_4^{2-}/EC ratio (3.42 ± 2.06). Only Toronto had a higher SO_4^{2-}/EC ratio (i.e., 4.93), mostly due to low EC levels. As noted earlier, there are spatial and seasonal variations in these ratios that reflect local and regional contributions.

 NO_3^{-}/SO_4^{2-} ratios have been used to evaluate relative contributions from coal-burning emissions, which abound in NO_x and SO_2 , and engine exhaust, which is a major NO_x emitter but contains little SO₂ (Hu et al., 2002; Wang et al., 2005; Yao et al., 2002). Average NO_3^{-}/SO_4^{2-} ratios were 0.61 in winter and 0.30 in summer. The NO_3^{-}/SO_4^{2-} ratio for Toronto (0.81) was >75% higher than the value found in this study (i.e., 0.46 ± 0.27), while the ratios in Seattle, WA (0.43), and Mexico City (0.45)were comparable. As/Fe and Pb/Fe ratios were 0.04 \pm 0.03 and 0.39 ± 0.32 , respectively, much higher than those for the other cities and indicative of the ash in uncontrolled coal combustion. Figure 3 shows a reasonably good association of Pb and SO_4^{2-} concentrations with the As marker for coal ash. The scatter (e.g., Figure 3d) is typical of different ash composition and SO₂ to SO_4^{2-} transformation rates. The Pb/As correlation indicates that the Pb more probably derives from the coal ash than from the remnants from leaded gasoline, as also indicated by differences in abundances for Pb isotopic ratios (Xu et al., 2012; Widory et al., 2010; Zheng et al., 2004).

Material balance

Material balances estimating organic matter and soil from their marker species are shown in Figures 4 and 5 for the winter and summer seasons. Consistent with the previous discussion, organic material (OM), SO_4^{2-} , NO_3^{-} , and NH_4^+ are large components. OC takes on an even larger role when its unmeasured hydrogen and oxygen components are taken into account as OM. The role of geological material is also enhanced when the Fe marker is leveraged by reasonable assumptions about its abundance in Chinese soils. Approximately 0 to 15% of the measured mass is not quantified by the chemical analysis, which is potentially due to unmeasured species, underestimations for weighting factors for OM and geological material, and uncertainties in filter equilibration and gravimetric analysis (Malm et al., 2011; Kajino et al., 2006).

For winter samples, contributions in order of importance were OM > geological material > sulfate > nitrate> ammonium > elemental carbon at major cities such as Beijing, Tianjin, Wuhan, Chongqing, Hangzhou, and Xiamen. Compositions

Table 2. Comparison of PM2.5 chemical component ratios for the 14 Chinese cities with ratios from selected cities in Europe, Canada, Mexico, and the United States

Cities	SO4 ²⁻ /OC	SO4 ²⁻ /EC	NO3 ⁻ /SO4 ²⁻	As/Fe	Pb/Fe	Reference
14 Chinese cites (winter and summer)	0.90 ± 0.43	3.42 ± 2.06	0.46 ± 0.27	0.04 ± 0.03	0.39 ± 0.32	This study
Over Europe	0.50	1.51	0.92	NA	NA	Schaap et al. (2004)
Toronto, ON, Canada	0.62	4.93	0.81	0.007	0.062	Lee et al. (2003)
Mexico City, Mexico	0.32	0.71	0.45	0.006	0.068	Vega et al. (2004)
Seattle, WA	0.45	1.13	0.43	0.015	0.099	Maykut et al. (2003)

Note: NA, data not available.

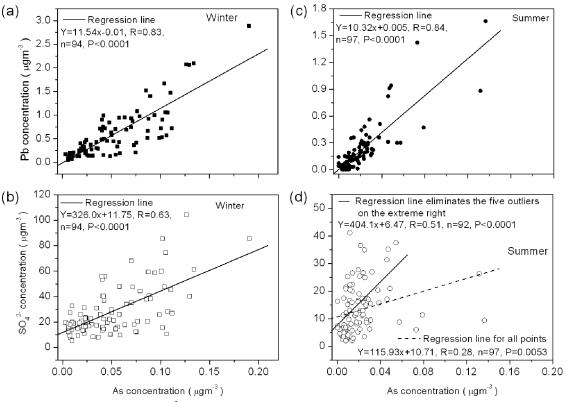


Figure 3. Relationships between $PM_{2.5}$ As, Pb, and SO_4^{2-} concentrations from the 14 cities during winter and summer, 2003.

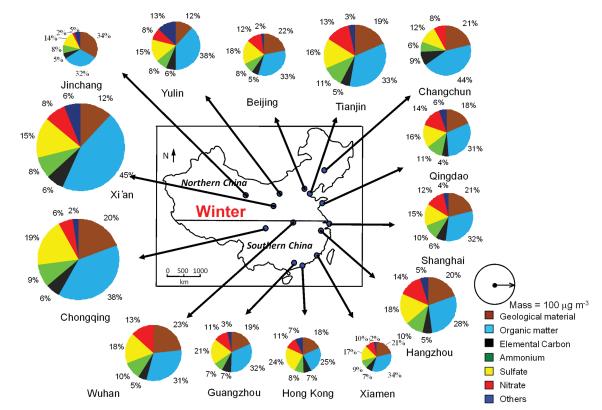


Figure 4. Wintertime material balance of $PM_{2.5}$ for the 14 Chinese cities. Organic matter (OM) is estimated as $1.6 \times OC$ (Chen and Yu, 2007; El-Zanan et al., 2005; El-Zanan et al., 2009) to account for unmeasured hydrogen and oxygen. Geological material is estimated as $25 \times Fe$ (Cao et al., 2008; Wu et al., 2011) to account for unmeasured oxygen and non-iron minerals. "Others" is the remaining unaccounted-for mass after subtracting the sum of measured components from the $PM_{2.5}$ mass. Unaccounted-for mass can be potentially composed of unmeasured geological material (e.g., calcium carbonate), a higher fraction of oxygen in OM, and liquid water associated with NH_4^+ , NO_3^- , and SO_4^{2-} at the 35% to 45% relative humidity filter weighing conditions.

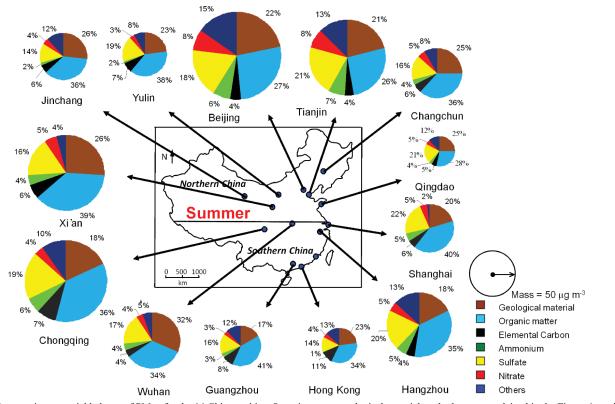


Figure 5. Summertime material balance of PM2.5 for the 14 Chinese cities. Organic matter, geological material, and others are explained in the Figure 4 caption.

differed for Yulin, Xi'an, and Hong Kong, where the wintertime SO_4^{2-} contribution exceeded that from geological material. At arid Jinchang, the geological material contribution exceeded the SO_4^{2-} and OM contributions.

During summer, most cities follow the general trend of OM > geological material > sulfate > nitrate, with elemental carbon contributions higher than ammonium contributions at all cities but Beijing and Tianjin. The contribution from SO_4^{2-} in Hong Kong differed between winter (25%) and summer (14%). At Shanghai and Hangzhou, SO_4^{2-} contributions exceeded those of geological material.

Comparison with other $PM_{2.5}$ speciation studies in Chinese cities

Table 3 compares city-specific results from this study with chemical concentrations from other major cities (i.e., Beijing, Xi'an, Shanghai, and Guangzhou). Although there are differences in magnitude owing to the differences in measurement periods, zones of representation, and measurement methods, the major components are similar in magnitude and order of importance for nearly all of the studies. There is no evidence of major upward or downward trends in mass and chemical composition from 1999 to 2006, but this is expected, given the short durations of the measurement programs and the large variability in emissions and meteorology expected over this time period. Trends in the United States have only been associated with emission reductions over long periods of a decade or more using chemically speciated measurements that are specific to those emissions.

PM_{2.5} in Chinese cities versus non-Chinese cities

Table 4 compares the major components from Chinese cities with $PM_{2.5}$ compositions in other countries. The geological material contribution is on the order of 10% in $PM_{2.5}$ from the non-Chinese cities, about half of that estimated from this study (19.5%). Roadside sites in St. Louis, MO, and Barcelona, Spain, showed more comparable geological material contributions (15.4 and 15.2%). The OM fractions in the Chinese cities are similar to most of the other cities, although the absolute OM concentrations are much higher in China. SO_4^{2-} , NO_3^{-} , and NH_4^+ are important in $PM_{2.5}$ in all of the cities, but their fractions are more variable and their absolute values are generally lower than those found in the 14 Chinese cities.

Average $PM_{2.5}$ concentrations for this study ranged from 3 to 9 times higher than the values in Seoul, Yokohama, St. Louis, Indianapolis, Toronto, Mexico City, Barcelona, and Milan, with corresponding 2 to 10 times higher levels of OM. The average Chinese secondary aerosol concentrations for SO_4^{2-} , NO_3^{-} , and NH_4^+ were 2–5, 1–10, and 2–7 times higher, respectively, than those in other cities in the world. As and Pb were 10 times and average geological material was 5–43 times those found in other cities.

Conclusions

 $PM_{2.5}$ in 14 of China's large cities achieved high concentrations in both winter and summer of 2003 with averages >100 µg m⁻³ being common occurrences. A grand average of 115 µg m⁻³ was

2012
September
25
at 20:10
te of Earth Environment]
[Institut
Downloaded by

GZ)	
G	
no	
zh	
gue	
Buž	
pi (
an	
Ê	
$\overline{\mathbf{s}}$	
hai	
ng	
Sha	
ı), Sh	
X	
0	
či'an (
X	
- 5	
e e	
in Beijing (
Beij	
nE	
di	
stu	
r PM _{2.5} stu	
PM	
er]	
oth	
Ĕ	
fron	
its	
nen	
ren	
nsu	
neź	
thr	
wit	
ĥ	
stud	
ı th	
from this	
n_3	
g D	
ਤ	
suo	
atic	
ntra	
cei	
con	
ale	
nic	
her	
r c	
ajo	
<u>п</u>	
anc	
5.5	
PM_2	
γf Ρ	
ų c	
iso	
par	
lmc	
ŭ	
3.	
ble	
Та	

						ć							
Location	Site description	Period (MM/YY)	PM _{2.5}	OC	EC	SO_4^{2-}	NO_3^-	$\mathrm{NH_4}^+$	\mathbf{K}^+	Е	As	Pb	Reference
Beijing (BJ)	Semiresidential (Tsinghua)	00/60-66/60	127.0	29.1	10.1	14.1	6.6	6.5	2.2	1.1	NA	0.34	He et al. (2001)
	Urban (Chegongzhuang)	00/60-66/60	115.0	21.5	8.7	14.5	10.3	6.2	2.2	1.1	NA	0.30	He et al. (2001)
	Semiresidential (Tsinghua)	08/01-09/02	106.9	28.8	9.6	10.6	7.8	5.5	1.4	1.2	0.06	0.21	Duan et al. (2006)
	Urban (Chegongzhuang)	08/01-09/02	96.6	22.9	10.3	9.6	6.9	5.8	1.7	1.1	0.06	0.17	Duan et al. (2006)
	Traffic (Beijing	2002–2003	106.5	22.4	8.1	23.2	14.6	11.7	1.6	0.8	0.03	0.21	Sun et al. (2004)
	Normal University)												
	Industrial	2002-2003	111.5	22.8	8.2	21.2	13.4	12.2	1.7	1.5	0.05	0.24	Sun et al. (2004)
	(Capital Steel Company)												
	Residential (Yihai Garden)	2002-2003	128.8	24.4	13.9	24.8	16.3	15.0	2.7	1.1	0.04	0.21	Sun et al. (2004)
	Urban	01/03, 06/03-07/03	123.6	21.8	5.9	21.3	13.4	9.6	1.5	0.9	0.03	0.23	This study
	Suburban	06/05-08/05	68.0	8.2	4.9	22.5	9.7	5.4	NA	NA	NA	NA	Pathak et al. (2011)
	Semiresidential	03/05-02/06	118.5	24.5	8.2	15.8	10.1	7.3	NA	1.1	0.02	0.24	Yang et al. (2011)
	(TSinghua)												
	and rural sites (Miyun)												
Xi'an (XA)	Urban	01/03, 06/03-07/03	229.6	60.2	14.0	35.3	17.0	17.0	3.3	1.4	0.08	1.22	This study
	Urban	09/03-02/04	199.4	48.0	13.3	NA	NA	NA	NA	NA	NA	NA	Cao et al. (2005)
	Urban	10/05-10/06	NA	NA	NA	19.5	8.3	8.7	1.8	NA	NA	NA	Shen et al. (2008)
	Urban	03/06-03/07	194.1	NA	NA	35.6	16.4	11.4	NA	NA	NA	NA	Zhang et al. (2011)
Shanghai (SH)	Urban	03/99-02/00	67.6	16.8	6.5	13.0	5.8	5.7	ΝA	0.9	NA	0.29	Ye et al. (2003)
	Urban	01/03, 06/03-07/03	96.7	20.0	6.1	16.8	10.1	9.0	1.5	0.8	0.03	0.32	This study
	Urban (Fudan and Taopu)	09/03-01/05	94.6	NA	NA	10.4	6.2	3.8	0.6	NA	NA	NA	Wang et al. (2006)
	Urban (Baoshan, Putuo,	04/04-05/04	NA	NA	NA	NA	NA	NA	NA	0.9	0.03	0.11	Chen et al. (2009)
	Huangpu, Jiading)												
	Urban	06/05-08/05	67.0	16.9	10.0	15.6	7.2	4.2	ΝA	NA	NA	ΝA	Pathak et al. (2011)
	Urban (Zhabei district)	10/05-07/06	90.3	14.7	2.8	NA	NA	NA	NA	NA	NA	NA	Feng et al. (2009)
	Subsurb (Jiading district)	10/05-07/06	95.5	17.5	3.0	NA	NA	NA	NA	NA	NA	NA	Feng et al. (2009)
Guangzhou (GZ)	Urban, industrial,	06/02-07/02	78.1	15.8	5.9	NA	NA	NA	NA	NA	NA	NA	Cao et al. (2004)
	and background mixed												
	Urban, industrial,	01/02-02/02	105.9	22.6	8.3	NA	NA	NA	NA	NA	NA	NA	Cao et al. (2004)
	and background mixed												
	Urban	10/02-06/03	70.6	17.6	4.4	14.7	4.0	4.5	NA	NA	NA	NA	Hagler et al. (2006)
	Urban	01/03, 06/03-07/03	75.0	17.1	5.8	13.5	6.3	4.8	1.4	0.5	0.03	0.27	This study
	Urban	10/04 - 11/04	102.9	22.4	7.1	27.8	4.3	12.1	2.6	NA	NA	NA	Andreae et al. (2008)
	Suburban	06/05-08/05	59.0	14.9	10.5	12.7	5.8	5.0	NA	NA	NA	NA	Pathak et al. (2011)
	Residential and commercial	12/08-02/09	81.7	17.5	4.1	5.6	12.0	4.7	NA	1.9	0.04	0.45	Yang et al. (2011)
	mixed (Wushan)												

Note: NA, data not available.

Table 4. Comparison of PM_{2.5} mass and major chemical components from this study with measurements from other major cities worldwide

						Cor	Concentration ($\mu g m^{-3}$)	$(\mu g m^{-3})$				
City	Site type	Period (MM/YY)	Mass	$\mathrm{SO_4}^{2-}$	NO_3^-	$\mathrm{NH_4}^+$	MO	EC	Geological matter	\mathbf{As}	Ъb	Reference
14 Cities, China	Urban	01/03, 06/03-07/03	115.2	19.89	9.98	9.20 (8.00/)	29.48	6.58	22.4	0.04	0.39	This study
Seoul, Korea	Urban	03/03-12/06	37.6	(1/.3%) 5.77	(8.7%) 5.17	(8.0%) 3.70	(%0.02) 11.36	()./%) 2.91	(19.5%)	NA	0.03	Heo et al. (2009)
				(15.3%)	(13.8%)	(9.8%)	(30.2%)	(0%.7%)	(7.4%)			~
Yokohama, Japan	Urban	09/07-08/08	20.6	3.80	0.96	2.27	4.50	1.94	NA	NA	NA	Khan et al. (2010)
				(18.4%)	(4.7%)	(11.0%)	(21.8%)	(9.4%)				
St.Louis, MO												
Blair	Traffic	02/00-12/03	16.4^{b}	4.23	2.48	1.94	3.25	0.84	2.53^{a}	0.003	0.013	Lee and Hopke (2006) ⁸⁷
				(25.8%)	(15.1%)	(11.8%)	(19.8%)	(5.1%)	(15.4%)			
Arnold	Traffic	02/00-12/03	15.5	4.16	1.9	1.61	3.3	0.58	0.52^a	0.005	0.019	
				(26.8%)	(12.4%)	(10.4%)	(21.3%)	(3.7%)	(3.4%)			
Indianapolis, IN	Urban	12/00-11/03	17.8^{a}	7.12	2.85	2.28	2.86	0.66	0.94^a	0.002	0.006	Zhao and Hopke (2006)
				(40.0%)	(16.0%)	(12.8%)	(16.0%)	(3.7%)	(5.3%)			
Toronto, Canada	Urban	02/00-02/01	12.7	2.27	1.84	1.19	4.39	0.46	1.38	0.0004	0.003	Lee et al. (2003)
				(17.9%)	(14.5%)	(9.4%)	(34.6%)	(3.6%)	(10.8)			
Mexico City, Mexico	Urban	2000-2002	35.05	5.90	2.67	2.95	21.88	8.35	2.62	0.01	0.12	Vega et al. (2004)
				(13.4%)	(5.9%)	(6.7%)	(48.9%)	(17.0%)	(3.5%)			
Northwestern Europe	Urban	1996–2007	NA	NA	NA	NA	NA	NA	NA	NA	NA	Putaud et al. (2010)
				(21%)	(16%)		(25%)	(0%)	(5%)			
Southern Europe	Urban	1996–2007										Putaud et al. (2010)
				(15%)	(%L)		(23%)	(8%)	(11%)			
Central Europe	Urban	1996–2007	NA	NA	NA	NA	NA	NA	NA	NA	NA	Putaud et al. (2010)
				(19%)	(13%)		(22%)	(14%)	(5%)			
Barcelona, Spain	Traffic	1999–2001	27.6	4.20	2.30	2.00	12.24	NA	4.20	NA	NA	Querol et al. (2004)
				(15.2%)	(8.3%)	(7.2%)	(44.3%)		(15.2%)			
Milan, Italy	Urban	08/02-12/03	40.0	4.70	8.70	3.00	15.48	1.40	1.50			Lonati et al. (2008)
				(11.8%)	(21.8%)	(7.5%)	(38.7%)	(3.5%)	(3.8%)			
Notes: Figures in the parentheses are the mass percentages of each chemical component. NA, data not available. ^a Estimated by mass balance model. ^b Median values.	theses are the r	nass percentages of each c	chemical c	omponent. N	lA, data not	available. ^a f	Estimated by	mass balan	ce model. ^b Median val	ues.		

Cao et al. / Journal of the Air & Waste Management Association 62 (2012) 1214–1226

found for all cities, with a minimum of 27.3 μ g m⁻³ measured at coastal Qingdao during summer and a maximum of 356.3 μ g m⁻³ at inland Xi'an during winter. Both primary and secondary PM2.5 are important contributors at all of the cities during both winter and summer. While ammonium sulfate is a large contributor during both seasons, ammonium nitrate contributions are much larger during winter. Lead levels are still high in several cities, reaching an average of 1.68 μ g m⁻³ in Xi'an during winter. High correlations of lead with arsenic and sulfate concentrations indicate that much of it derives from coal combustion rather than leaded fuels that were phased out by calendar year 2000. Although limited fugitive dust markers were available, scaling of iron by its ratios in source profiles shows $\sim 20\%$ of PM_{2.5} deriving from fugitive dust in most of the cities. Multipollutant control strategies will be needed that address incomplete combustion of coal and biomass, engine exhaust, and fugitive dust, as well as sulfur dioxide, oxide of nitrogen, and ammonia gaseous precursors for ammonium sulfate and ammonium nitrate.

Acknowledgments

This work was supported by the Natural Science Foundation of China (NSFC40925009), projects from Chinese Academy of Sciences (KZCX2-YW-BR-10, O929011018, and KZCX2-YW-148). The authors thank Jo Gerrard of the Desert Research Institute for her assistance in assembling and editing the paper.

References

- Andreae, M.O. 1983. Soot carbon and excess fine potassium: Long-range transport of combustion-derived aerosols. *Science* 220:1148–1151. doi:10.1126/ science.220.4602.1148
- Andreae, M.O., O. Schmid, H. Yang, D. Chand, J.Z. Yu, L.M. Zeng, and Y.H. Zhang. 2008. Optical properties and chemical composition of the atmospheric aerosol in urban Guangzhou, China. *Atmos. Environ.* 42(25):6335– 6350. doi:10.1016/j.atmosenv.2008.01.030
- Cao, J.J., F. Wu, J.C. Chow, S.C. Lee, Y. Li, S.W. Chen, Z.S.An, K.K., Fung, J.G. Watson, C.S. Zhu, and S.X. Liu. 2005. Characterization and source apportionment of atmospheric organic and elemental carbon during fall and winter of 2003 in Xi'an, China. *Atmos. Chem. Phys.* 5:3127–3137. 1680-7324/acp/ 2005-5-3127 doi:10.5194/acp-5-3127-2005
- Cao, J. J., H. Li, J. C. Chow, J.G. Watson, S.C. Lee, B. Rong, J.G. Dong, and K.F. Ho. 2011. Chemical composition of indoor and outdoor atmospheric particles at Emperor Qin's terra-cotta museum, Xi'an, China. *Aerosol Air Qual. Res.* 11(1):70–79. doi:10.4209/aaqr.2010.10.0088
- Cao, J.J., J.C. Chow, J.G. Watson, F. Wu, Y.M. Han, Z.D. Jin, Z.X. Shen, and Z. S. An. 2008. Size-differentiated source profiles for fugitive dust in the Chinese Loess Plateau. *Atmos. Environ.* 42(10):2261–2275. doi:10.1016/j.atmosenv. 2008.09.043
- Cao, J.J., J.C. Chow, J. Tao, S.C. Lee, J.G. Watson, K.F. Ho, G.H. Wang, C.S. Zhu, and Y.M. 2011. Han. Stable carbon isotopes in aerosols from Chinese cities: Influence of fossil fuels. *Atmos. Environ.* 45(6):1359–1363. doi:10.1016/j. atmosenv.2010.10.056
- Cao, J.J., S.C. Lee, J.C. Chow, J.G. Watson, K.F. Ho, R.J. Zhang, Z.D. Jin, Z.X. Shen, G.C. Chen, Y.M. Kang, S.C. Zou, L.Z. Zhang, S.H. Qi, M.H. Dai, Y. Cheng, and K. Hu. 2007. Spatial and seasonal distributions of carbonaceous aerosols over China. J. Geophys. Res. Atm. 112(D22S11):1–9. doi:10.1029/ 2006JD008205
- Cao, J.J., S.C. Lee, K.F. Ho, S.C. Zou, K.K. Fung, Y. Li, J.G. Watson, and J.C. Chow. 2004. Spatial and seasonal variations of atmospheric organic carbon and elemental carbon in Pearl River Delta Region, China. *Atmos. Environ.* 38:4447–4456. doi:10.1016/j.atmosenv.2004.05.016

- Cao, J.J., S.C. Lee, K.F. Ho, X.Y. Zhang, S.C. Zou, K.K. Fung, J.C. Chow, and J.G. Watson. 2003. Characteristics of carbonaceous aerosol in Pearl River Delta region, China during 2001 winter period. *Atmos. Environ.* 37 (11):1451–1460. doi:10.1016/S1352-2310(02)01002-6
- Chan, C.K., and X.Yao. 2008. Air pollution in mega cities in China. Atmos. Environ. 42(1):1–42. doi:10.1016/j.atmosenv.2007.09.003
- Chen, J.M., M.G. Tan, Y.L. Li, J. Zheng, Y.M. Zhang, Z. Shan, G.L. Zhang, and Y. Li. 2008. J. Hazard. Mater. 156:36–43. doi:10.1016/j.jhazmat.2007.11.122
- Chen, X., and J.Z. Yu. 2007. Measurement of organic mass to organic carbon ratio in ambient aerosol samples using a gravimetric technique in combination with chemical analysis. *Atmos. Environ.* 41:8857–8864. doi:10.1016/j. atmosenv.2007.08.023
- Chow, J.C., and J.G. Watson. 2012. Chemical analyses of particle filter deposits. In Aerosols Handbook: Measurement, Dosimetry, and Health Effects. Ed. L. Ruzer and N.H. Harley. New York, NY: CRC Press/Taylor & Francis.
- Chow, J.C., and J.G.Watson. 2011. Air quality management of multiple pollutants and multiple effects. Air Qual. Climate Change J. 45(3):26–32.
- Chow, J.C., J.G. Watson, J. Robles, X.L. Wang, L.-W.A. Chen, D.L. Trimble, S. D. Kohl, R.J. Tropp, and K.K. Fung. 2011. Quality assurance and quality control for thermal/optical analysis of aerosol samples for organic and elemental carbon. *Anal. Bioanal. Chem.* 401(10):3141–3152. doi: 10.1007/ s00216-011-5103-3
- Chow, J.C., J.G. Watson, L.C. Pritchett, W.R. Pierson, C.A. Frazier, and R.G. Purcell. 1993. The DRI thermal/optical reflectance carbon analysis system: Description, evaluation and applications in U.S. air quality studies. *Atmos. Environ.* 27A(8):1185–1201. doi:10.1016/0960-1686(93)90245-T
- Chow, J.C., J.G. Watson, L.-W.A. Chen, M.C.O. Chang, N.F. Robinson, D.L. Trimble, and S.D. Kohl. 2007. The IMPROVE_A temperature protocol for thermal/optical carbon analysis: Maintaining consistency with a long-term database. J. Air Waste Manage. Assoc. 57(9):1014–1023. doi:10.3155/1047-3289.57.9.1014
- Chow, J.C., J.G. Watson, L.-W.A. Chen, S.S.H. Ho, D. Koracin, B. Zielinska, D. Tang, F. Perera, J.J. Cao, and S.C. Lee. 2006. Exposure to PM_{2.5} and PAHs from the Tong Liang, China, epidemiological study. J. Environ. Sci. Health Part A A41(4):517–542. doi:10.1080/10934520600564253
- Chow, J.C., J.P. Engelbrecht, J.G. Watson, W.E. Wilson, N.H. Frank, and T. Zhu. 2002. Designing monitoring networks to represent outdoor human exposure. *Chemosphere* 49(9):961–978. doi:10.1016/S0045-6535(02)00239-4
- Deng, C., G. Zhuang, K. Huang, J. Li, R. Zhang, Q. Wang, T. Liu, Y. Sun, Z. Guo, J.S. Fu, and Z. Wang. 2011. Chemical characterization of aerosols at the summit of Mountain Tai in Central East China. *Atmos. Chem. Phys.* 11 (14):7319–7332. doi:10.5194/acp-11-7319-2011
- Du, K., W. Yuen, W. Wang, M.J. Rood, R.M. Varma, R.A. Hashmonay, B.J. Kim, and M.R. Kemme. 2011. Optical remote sensing to quantify fugitive particulate mass emissions from stationary short-term and mobile continuous sources: Part II. Field applications. *Environ. Sci. Technol.* 45(2):666–672. doi:10.1021/es101906v
- Duan, F.K., K.B. He, Y.L. Ma, F.M. Yang, X.C. Yu, S.H. Cadle, T. Chan, and P. A. Mulawa. 2006. Concentration and chemical characteristics of PM_{2.5} in Beijing, China: 2001–2002. *Sci. Total Environ.* 355(1–3): 264–275. doi:10.1016/j.scitotenv.2005.03.001
- Duan, F.K., X.D. Liu, T. Yu, and H. Cachier. 2004. Identification and estimate of biomass burning contribution to the urban aerosol organic carbon concentrations in Beijing. *Atmos. Environ.* 38(9):1275–1282. doi:10.1016/j. scitotenv.2005.03.001
- El-Zanan, H.S., B. Zielinska, L.R. Mazzoleni, and D.A. Hansen. 2009. Analytical determination of the aerosol organic mass-to-organic carbon ratio. J. Air Waste Manage. Assoc. 59(1):58–69. doi:10.3155/1047-3289.59.1.58
- El-Zanan, H.S., D.H. Lowenthal, B. Zielinska, J.C. Chow, and N.K. Kumar. 2005. Determination of the organic aerosol mass to organic carbon ratio in IMPROVE samples. *Chemosphere* 60(4):485–496. doi:10.1016/j. chemosphere.2005.01.005
- Feng, Y.L., Y.J. Chen, H. Guo, G.R. Zhi, S.C. Xiong, J. Li, G.Y. Sheng, and J.M. Fu. 2009. Characteristics of organic and elemental carbon in PM_{2.5} samples in Shanghai, China. *Atmos. Res.* 92(4):434–442. doi:10.1016/j. atmosres.2009.01.003

- Gong, S.L., and X.Y. Zhang. 2008. CUACE/Dust—An integrated system of observation and modeling systems for operational dust forecasting in Asia. *Atmos. Chem. Phys.* 8(9):2333–2340. doi:10.5194/acp-8-2333-2008
- Gu, J.X., Z.P. Bai, W.F. Li, L.P. Wu, A.X. Liu, H.Y. Dong, and Y.Y. Xie. 2011. Chemical composition of PM(2.5) during winter in Tianjin, China. *Particuology* 9(3):215–221. doi:10.1016/j.partic.2011.03.001
- Guinot, B., H. Cachier, J. Sciare, Y. Tong, W. Xin, and Y. Jianhua. 2007. Beijing aerosol: Atmospheric interactions and new trends. J. Geophys. Res. Atmos. 112(D14). doi:10.1029/2006JD008195
- Hagler, G.S. W., M.H. Bergin, L.G. Salmon, J.Z. Yu, E.C.H. Wan, M. Zheng, L.M. Zeng, C.S. Kiang, Y.H. Zhang, A.K.H. Lau, and J.J. Schauer. 2006. Source areas and chemical composition of fine particulate matter in the Pearl River Delta region of China. *Atmos. Environ.* 40(20):3802–3815. doi:10.1016/j.atmosenv.2006.02.032
- Han, J., and Y. Hayashi. 2008. Assessment of private car stock and its environmental impacts in China from 2000 to 2020. *Transport. Res. Part D Transport Environ.* 13(7):471–478. doi:10.1016/j.trd.2008.09.007
- Han, Y.M., S.C. Lee, J.J. Cao, K.F. Ho, and Z.S. An. 2009. Spatial distribution and seasonal variation of char-EC and soot-EC in the atmosphere over China. *Atmos. Environ.* 43(38):6066–6073. doi:10.1016/j.atmosenv.2009.08.018
- He, K.B., F. Yang, Y. Ma, Q. Zhang, X. Yao, C.K. Chan, S.H. Cadle, T. Chan, and P. Mulawa. 2001. The characteristics of PM_{2.5} in Beijing, China. *Atmos. Environ.* 35(29):4959–4970. doi:10.1016/S1352-2310(01)00301-6
- Heo, J.B., P.K. Hopke, and S.M. Yi. 2009. Source apportionment of PM_{2.5} in Seoul, Korea. *Atmos. Chem. Phys.* 9(14): 4957–4971. doi:10.5194/acp-9-4957-2009
- Ho, K.F., J.J. Cao, S.C. Lee, K. Kawamura, R.J. Zhang, J.C. Chow, and J.G. Watson. 2007. Dicarboxylic acids, ketocarboxyylic acids, and dicarbonyls in the urban atmosphere of China. *Journal of Geophysical Research-Atmospheres*. D22(S27): S27-1–S27-12. doi:10.1029/2006J008011
- Ho, K.F., S.C. Lee, J.J. Cao, J.C. Chow, J.G. Watson, and C.K. Chan. 2006. Seasonal variations and mass closure analysis of particulate matter in Hong Kong. *Sci. Total Environ.* 355(1–3):276–287. doi:10.1016/j.scitotenv. 2005.03.013
- Hu, D., Q.J. Bian, A.K.H. Lau, and J.Z. Yu. 2010. Source apportioning of primary and secondary organic carbon in summer PM2.5 in Hong Kong using positive matrix factorization of secondary and primary organic tracer data. J. Geophys. Res. Atmos. 115(D16). doi:10.1029/2009JD012498
- Hu, M., L.Y. He, Y.H. Zhang, M. Wang, Y.P. Kim, and K.C. Moon. 2002. Seasonal variation of ionic species in fine particles at Qingdao, China. *Atmos. Environ.* 36(38):5853–5859. doi:10.1016/S1352-2310(02)00581-2
- Hu, T.F., S.C. Lee, J.J. Cao, W.K. Ho, K.F. Ho, J.C. Chow, J.G. Watson, B. Rong, and Z.S. An. 2009. Atmospheric deterioration of Qin brick in an environmental chamber at Emperor Qin's Terracotta Museum, China. *Journal of Archaeological Science* 36(11):2578–2583. doi:10.1016/j.jas.2009.07.014
- Kajino, M., W. Winwarter, and H. Ueda. 2006. Modeling retained water content in measured aerosol mass. *Atmos. Environ.* 40(27):5202–5213. doi:10.1016/ j.atmosenv.2006.04.016
- Khan, M. F., Y. Shirasuna, K. Hirano, and S. Masunaga. 2010. Characterization of PM_{2.5}, PM_{2.5-10} and PM>10 in ambient air, Yokohama, Japan. *Atmos. Res.* 96(1):159–172. doi:10.1016/j.atmosres.2009.12.009
- Lee, J.H., and P.K. Hopke. 2006. Apportioning sources of PM_{2.5} in St. Louis, MO using speciation trends network data. *Atmos. Environ.* 40(suppl. 2):S360– S377. doi:10.1016/j.atmosenv.2005.11.074
- Lee, P.K.H., J.R. Brook, E. Dabek-Zlotorzynska, and S.A. Mabury. 2003. Identification of the major sources contributing to PM_{2.5} observed in Toronto. *Environ. Sci. Technol.* 37(21):4831–4840. doi:10.1021/es026473i
- Li, X.X., J.J. Cao, J.C. Chow, Y.M. Han, S.C. Lee, and J.G. Watson. 2008. Chemical characteristics of carbonaceous aerosols during dust storms over Xi'an in China. Adv. Atmos. Sci. 25(5):847–855. doi:10.1007/s00376-008-0847-1
- Lighty, J.S., J.M. Veranth, and A.F. Sarofim. 2000. Critical review: Combustion aerosols: Factors governing their size and composition and implications to human health. J. Air Waste Manage. Assoc. 50(9):1565–1618. doi:10.1080/ 10473289.2000.10464197

- Lonati, G., M. Giugliano, and S. Ozgen. 2008. Primary and secondary components of PM_{2.5} in Milan (Italy). *Environ. Int.* 34(5):665–670. doi:10.1016/j. envint.2007.12.009
- Louie, P.K.K., J.C. Chow, L.-W.A. Chen, J.G. Watson, G. Leung, and D. Sin. 2005. PM_{2.5} chemical composition in Hong Kong: Urban and regional variations. *Sci. Total Environ.* 338(3):267–281. doi:10.1016/j.scitotenv. 2004.07.021
- Louie, P.K.K., J.G. Watson, J.C. Chow, L.-W.A. Chen, D.W.M. Sin, and A.K.H. Lau. 2005. Seasonal characteristics and regional transport of PM_{2.5} in Hong Kong. *Atmos. Environ.* 39(9):1695–1710. doi:10.1016/j.atmosenv.2004. 11.017
- Malm, W.C., B.A. Schichtel, and M.L. Pitchford. 2011. Uncertainties in PM(2.5) gravimetric and speciation measurements and what we can learn from them. J. Air Waste Manage. Assoc. 61(11):1131–1149. doi:10.1080/10473289. 2011.603998
- Mauderly, J.L., and J.C. Chow. 2008. Health effects of organic aerosols. *Inhal. Toxicol.* 20(3):257–288. doi:10.1080/08958370701866008
- Maykut, N.N., J. Lewtas, E. Kim, and T.V. Larson. 2003. Source apportionment of PM_{2.5} at an urban IMPROVE site in Seattle, Washington. *Environ. Sci. Technol.* 37(22):5135–5142. doi:10.1021/es030370y
- NAE, NRC, CAE, and CAS. 2008. Energy Futures and Urban Air Pollution Challenges for China and the United States. Washington, DC: National Academies Press.
- Pandis, S.N., J.H. Seinfeld, and C. Pilinis. 1992. Heterogeneous sulfate production in an urban fog. *Atmos. Environ.* 26A(14):2509–2522. doi:10.1016/ 0960-1686(92)90103-R
- Pathak, R.K., T. Wang, K.F. Ho, and S.C. Lee. 2011. Characteristics of summertime PM_{2.5} organic and elemental carbon in four major Chinese cities: Implications of high acidity for water-soluble organic carbon (WSOC). *Atmos. Environ.* 45(2):318–325. doi:10.1016/j.atmosenv.2010.10.021
- Pope, C.A. III, and D.W. Dockery. 2006. Critical Review: Health effects of fine particulate air pollution: Lines that connect. J. Air Waste Manage. Assoc. 56 (6):709–742. doi:10.1080/10473289.2006.10464485
- Putaud, J.P., R. van Dingenen, A. Alastuey, H. Bauer, W. Birmili, J. Cyryse, H. Flentje, S. Fuzzi, R. Gehrigh, H.C. Hansson, R.M. Harrison, H. Herrmann, R. Hitzenberger, C. Huglin, A.M. Jones, A. Kasper-Giebl, G. Kiss, A. Kousa, T.A.J. Kuhlbusch, G. Loschau, W. Maenhaut, A. Molnar, T. Moreno, J. Pekkanen, C. Perrino, M. Pitz, H. Puxbaum, X. Querol, S. Rodriguez, I. Salma, J. Schwarz, J. Smolik, J. Schneider, G. Spindler, H. Ten Brink, J. Tursic, M. Viana, A. Wiedensohler, and F.A. Raes. 2010. European aerosol phenomenology—3: Physical and chemical characteristics of particulate matter from 60 rural, urban, and kerbside sites across Europe. *Atmos. Environ.* 44(10):1308–1320. doi:10.1016/j.atmosenv.2009.12.011
- Pytkowicz, R.M., and D.R. Kester. 1971. The physical chemistry of sea water. Oceanogr. Mar. Biol. 9:11–60. doi:10.1016/0011-7471(71)90058-1
- Qian, Y., F. Giorgi, Y. Huang, W.L. Chameides, and C. Luo. 2001. Regional simulation of anthropogenic sulfur over East Asia and its sensitivity to model parameters. *Tellus Seri. B Chem. Phys. Meteorol.* 53(2):171–191. doi:10.1016/0011-7471(71)90058-1
- Querol, X., A. Alastuey, M.M. Viana, S. Rodriguez, B. Artinano, P. Salvador, S.G. do Santos, R.F. Patier, C.R. Ruiz, J. de la Rosa, A.S. de la Campa, M. Menendez, and J.I. Gil. 2004. Speciation and origin of PM₁₀ and PM_{2.5} in Spain. J. Aerosol Sci. 35(9):1151–1172. doi:10.1016/j.jaerosci.2004.04.002
- Schaap, M., H.A.C.D. van der Gon, F.J. Dentener, A.J.H. Visschedijk, M. Van Loon, H.M. ten Brink, J.P. Putaud, B. Guillaume, C. Liousse, and P.J.H. Builtjes. 2004. Anthropogenic black carbon and fine aerosol distribution over Europe. J. Geophys. Res. Atm. 109(D18). doi:10.1029/2003JD004330
- Shen, Z.X., J.J. Cao, R. Arimoto, R.J. Zhang, D.M. Jie, S.X. Liu, and C.S. Zhu. 2007. Chemical composition and source characterization of spring aerosol over Horqin sand land in northeastern China. J. Geophys. Res. Atm. 112 (D14). doi:10.1029/2006JD007991
- Shen, Z.X., J.J. Cao, R. Arimoto, Y.M. Han, C.S. Zhu, J. Tian, and S.X. Liu. 2010. Chemical characteristics of fine particles (PM₁) from Xi'an, China. *Aerosol Sci. Technol.* 44(6):461–472. doi:10.1080/02786821003738908
- Shen, Z.X., J.J. Cao, R. Arimoto, Z.W. Han, R.J. Zhang, Y.M. Han, S.X. Liu, T. Okuda, S. Nakao, and S. Tanaka. 2009. Ionic composition of TSP and PM_{2.5}

during dust storms and air pollution episodes at Xi'an, China. *Atmos. Environ.* 43(18):2911–2918. doi:10.1016/j.atmosenv.2009.03.005

- Shen, Z.X., R. Arimoto, J.J. Cao, R.J. Zhang, X.X. Li, N. Du, T. Okuda, S. Nakao, and S. Tanaka. 2008. Seasonal variations and evidence for the effectiveness of pollution controls on water soluble inorganic species in total suspended particulates and fine particulate matter from Xi'an, China. J. Air Waste Manage. Assoc. 58(12):1560–1570. doi:10.3155/1047-3289.58.12.1560
- So, K.L., H. Guo, and Y.S. Li. 2007. Long-term variation of PM_{2.5} levels and composition at rural, urban, and roadside sites in Hong Kong: Increasing impact of regional air pollution. *Atmos. Environ.* 41(40):9427–9434.
- Song, Y., X. Y. Tang, S.D. Xie, Y.H. Zhang, Y. J. Wei, M.S. Zhang, L.M. Zeng, and S.H. Lu. 2007. Source apportionment of PM_{2.5} in Beijing in 2004. *J. Hazard. Mater.* 146(1–2):124–130.
- Stelson, A.W., S.K. Friedlander, and J.H. Seinfeld. 1979. A note on the equilibrium relationship between ammonia and nitric acid and particulate ammonium nitrate. *Atmos. Environ.* 13:369–371. doi:10.1016/0004-6981(79)90293-2
- Sun, Y.L., G.S. Zhuang, W. Ying, L.H. Han, J.H. Guo, D. Mo, W.J. Zhang, Z.F. Wang, and Z.P. Hao. 2004. The air-borne particulate pollution in Beijing— Concentration, composition, distribution and sources. *Atmos. Environ.* 38 (35):5991–6004. doi:10.1016/j.atmosenv.2004.07.009
- Tian, H.Z., Y. Wang, Z.G. Xue, Y.P. Qu, F.H. Chai, and J.M. Hao. 2011. Atmospheric emissions estimation of Hg, As, and Se from coal-fired power plants in China, 2007. *Sci. Total Environ.* 409(16):3078–3081. doi:10.1016/j. scitotenv.2011.04.039
- Tie, X.X. and J.J. Cao. 2009. Aerosol pollution in China: Present and future impact on environment. *Particuology*. 7(6):426–431. doi:10.1016/j. partic.2009.09.003
- Vega, E., E. Reyes, H. Ruiz, J. Garcia, G. Sanchez, G. Martinez-Villa, U. Gonzalez, J.C. Chow, and J.G. Watson. 2004. Analysis of PM_{2.5} and PM₁₀ in the atmosphere of Mexico City during 2000–2002. *J. Air Waste Manage. Assoc.* 54(7):786–798. doi:10.1080/10473289.2004.10470952
- Wang, G.H., K. Kawamura, S. Lee, K.F. Ho, and J.J. Cao. 2006. Molecular, seasonal, and spatial distributions of organic aerosols from fourteen Chinese cities. *Environ. Sci. Technol.* 40(15):4619–4625. doi:10.1021/es060291x
- Wang, M.S., B.S. Zheng, B.B. Wang, L.A.H. Shehong, D.S. Wu, and J. Hu. 2006. Arsenic concentrations in Chinese coals. *Sci. Total Environ.* 357(1–3): 96–102. doi:10.1016/j.scitotenv.2005.04.045
- Wang, X.H., X.H. Bi, G.Y. Sheng, and J.M. Fu. 2006. Chemical composition and sources of PM₁₀ and PM_{2.5} aerosols in Guangzhou, China. *Environ. Monit. Assess.* 119(1–3):425–439. doi:10.1007/s10661-005-9034-3
- Wang, Y., G.S. Zhuang, A.H. Tang, H. Yuan, Y.L. Sun, S.A. Chen, and A.H. Zheng. 2005. The ion chemistry and the source of PM_{2.5} aerosol in Beijing. *Atmos. Environ.* 39(21):3771–3784. doi:10.1016/j.atmosenv.2005.03.013
- Wang, Y., G.S. Zhuang, C. Xu, and Z.S. An. 2007. The air pollution caused by the burning of fireworks during the lantern festival in Beijing. *Atmos. Environ*. 41 (2):417–431. doi:10.1016/j.atmosenv.2006.07.043
- Wang, Y., G. Zhuang, X. Zhang, K. Huang, C. Xu, A. Tang, J. Chen, and Z. An. 2006. The ion chemistry, seasonal cycle, and sources of PM2.5 and TSP aerosol in Shanghai. *Atmos. Environ.* 40:2935–2952. doi:10.1016/j. atmosenv.2005.12.051
- Watson, J.G. 2002. Visibility: Science and regulation—2002 Critical review. J. Air Waste Manage. Assoc. 52(6):628–713. doi:10.1080/10473289.2002.10470813
- Watson, J.G., J.C. Chow, L.-W.A. Chen, S.D. Kohl, and G. Casuccio. 2012. Elemental analysis of filter tape deposits from beta attenuation monitors. *Atmos. Res.* 106:181–189. doi:10.1016/j.atmosres.2011.12.004
- Widory, D., X.D. Liu, and S.P. Dong. 2010. Isotopes as tracers of sources of lead and strontium in aerosols (TSP & PM_{2.5}) in Beijing. *Atmos. Environ.* 44 (30):3679–3687. doi:10.1016/j.atmosenv.2010.06.036
- Wu, F., J.C. Chow, Z.S. An, J.G. Watson, and J.J. Cao. 2011. Size-differentiated chemical characteristics of Asian Paleo dust: Records from aeolian deposition on the Chinese loess plateau. J. Air Waste Manage. Assoc. 61(2):180– 189. doi:10.3155/1047-3289.61.2.180
- Wu, Y., J.M. Hao, L.X. Fu, J.N. Hu, Z.S. Wang, and U. Tang. 2003. Chemical characteristics of airborne particulate matter near major roads and at background locations in Macao, China. *Sci. Total Environ.* 317(1–3):159–172. doi:10.1016/S0048-9697(03)00331-0

- Xu, H.M., J.J. Cao, K.F. Ho, H. Ding, Y.M. Han, G.H. Wang, J.C. Chow, J.G. Watson, S.D. Kohl, J. Qiang, and W.T. Li. 2012. Lead concentrations in fine particulate matter after the phasing out of leaded gasoline in Xi'an, China. *Atmos. Environ.* 46:217–114. doi:10.1016/j.atmosenv.2011.09.078
- Xu, J., M.H. Bergin, R. Greenwald, J.J. Schauer, M.M. Shafer, J.L. Jaffrezo, and G. Aymoz. 2004. Aerosol chemical, physical, and radiative characteristics near a desert source region of northwest China during ACE-Asia. J. Geophys. Res. Atm. 109(D19). doi:10.1029/2003JD004239
- Xuan, J., I.N. Sokolik, J.F. Hao, F.H. Guo, H.Q. Mao, and G.M. Yang. 2004. Identification and characterization of sources of atmospheric mineral dust in East Asia. *Atmos. Environ.* 38(36):6239–6252. ISI:000225010500019. doi:10.1016/j.atmosenv.2004.06.042
- Yan, X.Y., T. Ohara, and H. Akimoto. 2006. Bottom-up estimate of biomass burning in mainland China. Atmos. Environ. 40(27):5262–5273. doi:10.1016/ j.atmosenv.2006.04.040
- Yang, F., J. Tan, Q. Zhao, Z. Du, K.B. He, Y. Ma, F. Duan, G. Chen, and Q. Zhao. 2011. Characteristics of PM_{2.5} speciation in representative megacities and across China. *Atmos. Chem. Phys.* 11(11):5207–5219. doi:10.5194/acp-11-5207-2011
- Yao, X., C.K. Chan, M. Fang, S.H. Cadle, T. Chan, P. Mulawa, K.B. He, and B. Ye. 2002. The water-soluble ionic composition of PM_{2.5} in Shanghai and Beijing, China. *Atmos. Environ.* 36(26):4223–4234. doi:10.1016/S1352-2310(02)00342-4
- Ye, B.M., X.L. Ji, H.Z. Yang, X.H. Yao, C.K. Chan, S.H. Cadle, T. Chan, and P. A. Mulawa. 2003. Concentration and chemical composition of PM_{2.5} in Shanghai for a 1-year period. *Atmos. Environ.* 37(4):499–510. doi:10.1016/ S1352-2310(02)00918-4
- Zhang, T., J.J. Cao, X.X. Tie, Z.X. Shen, S.X. Liu, H. Ding, Y.M. Han, G.H. Wang, K.F. Ho, J. Qiang, and W.T. Li. 2011. Water-soluble ions in atmospheric aerosols measured in Xi'an, China: Seasonal variations and sources. *Atmos. Res.* 102(1–2):110–119. doi:10.1016/j.atmosres.2011.06.014
- Zhang, W.J., W. Wang, J.H. Chen, H.J. Liu, T.Y. Dai, X.Y. Yang, F. Zhang, J. Lin, and Z.F. Wang. 2010. Pollution situation and possible markers of different sources in the Ordos Region, Inner Mongolia, China. *Sci. Total Environ.* 408 (3):624–635.
- Zhang, Z.Q., and S.K. Friedlander. 2000. A comparative study of chemical databases for fine particle Chinese. *Environ. Sci. Technol.* 34(22):4687–4694.
- Zhao, Q., K.B. He, K.A. Rahn, Y. Ma, Y. Jia, F. Yang, F. Duan, Y. Lei, G. Chen, Y. Cheng, H. Liu, and S. Wang. 2010. Dust storms come to Central and Southwestern China, too: Implications from a major dust event in Chongqing. *Atmos. Chem. Phys.* 10(6):2615–2630. doi:10.5194/acp-10-2615-2010
- Zhao, W.X., and P.K. Hopke. 2006. Source investigation for ambient PM_{2.5} in Indianapolis, IN. Aerosol Sci. Technol. 40(10):898–909. doi:10.1080/ 02786820500380297
- Zheng, J., M.G. Tan, Y. Shibata, A. Tanaka, Y. Li, G.L. Zhang, Y.M. Zhang, and Z. Shan. 2004. Characteristics of lead isotope ratios and elemental concentrations in PM₁₀ fraction of airborne particulate matter in Shanghai after the phase-out of leaded gasoline. *Atmos. Environ.* 38(8):1191–1200. doi:10.1016/j.atmosenv.2003.11.004

About the Authors

Jun-Ji Cao, Kin-Fai Ho, Ge-Hui Wang, and Yong-Ming Han are professors in the Division of Aerosol & Environment, Institute of Earth Environment, Chinese Academy of Sciences.

Xue-Xi Tie is a scientist at National Center for Atmospheric Research, USA.

Zhen-Xing Shen is an associate professor at Xi'an Jiaotong University, China.

Judith C. Chow and John G. Watson are research professors in Desert Research Institute, USA.

Shun-Cheng Lee is a professor in the Hong Kong Polytechnic University, Hong Kong.